

Published in:

Analytical and Bioanalytical Chemistry (2017), 409:1815-1825

The final publication is available at Springer via <https://doi.org/10.1007/s00216-016-0126-4>

4 Origins of the difference between food folate analysis by LC-MS/MS and
5 microbiological assay and quantitation of an oxidation product of 5-CH₃-
6 H₄folate

7 Christiane Ringling¹ and Michael Rychlik^{1,2 *}

8 ¹ Division BIOANALYTIK Weihenstephan, Research Center for Nutrition and
9 Food Sciences, Technische Universität München, Alte Akademie 10, D-
10 85350 Freising, Germany

11 ² Chair of Analytical Food Chemistry, Technische Universität München, Alte
12 Akademie 10, D-85350 Freising, Germany

13

14 **Keywords:** LC-MS/MS; food folate; stable isotope dilution assay;
15 microbiological assay; oxidation product of 5-CH₃-H₄folat;
16 pyrazino-s-triazine; MeFox

17 **Running head:** LC-MS/MS vs. Microbiolocial assay

18 Phone +49-8161-71 3746

19 Fax +49-8161-71 4216

20 E-mail michael.rychlik@tum.de

21

22 * Author to whom correspondence should be addressed

1 **ABSTRACT**

2 To investigate the often reported disagreement in food folate quantitation between the
3 microbiological assay and HPLC methods, different foods were analyzed both by the
4 microbiological assay and an LC-MS/MS method in the present study.

5 For the LC-MS/MS analysis we emphasize the need for a complete deconjugation of
6 polyglutamic folate forms. Moreover, our results revealed no necessity for an additional
7 enzyme treatment except from the deconjugation step. To check the efficiency of
8 deconjugation without additional sample preparations a quantification of the amount of
9 diglutamates and a screening for additional polyglutamates was applied. An intense
10 investigation of a substance with a polyglutamate chain being deconjugated like the
11 folates revealed that it was an oxidation product of 5-CH₃-H₄folate, a pyrazino-s-triazine
12 called MeFox in previous reports. The latter is not microbiologically active and, therefore,
13 does not contribute to the amount of total folates. But we found it being commonly present
14 in foods, especially in those being low in ascorbic acid.

15 The microbiological assay showed different responses to the single vitamers. Therefore, it
16 was necessary to calibrate with the folate, which had the highest portion of the folate
17 distribution.

18 The investigations showed that both methods can provide similar results, when including
19 for both a deconjugation step. This is particularly important for LC-MS/MS but probably
20 also for the microbiological assay. Additionally, considering the folate distribution was
21 found to be crucial for the accurate calibration of the microbiological assay.

22

1 1. Introduction

2 Folates are a group of water-soluble vitamins playing a key role in C₁-metabolism. For
3 their analysis, a microbiological assay and HPLC-methods with various detectors (i.a. UV,
4 Fluorescence and MS) are commonly applied. Whereas the microbiological assay is very
5 sensitive but reveals only the total amount of folates the HPLC allows to differentiate and
6 quantitate the single vitamers. However, in particular for HPLC attention must be paid to a
7 good sample clean up and to a complete deconjugation to the respective folate
8 monoglutamates. But even for the microbiological assay a certain degree of deconjugation
9 is necessary, because the mostly applied microorganism *Lactobacillus rhamnosus* has a
10 similar response to mono-, di- and triglutamates, but a decreasing activity with further
11 increasing chain length [1].

12 For both methodologies, an extraction with three different kinds of enzymes is
13 recommended [2], the so called “trienzyme treatment” introduced by Martin et al. [3]. Here,
14 in addition to the enzymes for deconjugation, additional treatment with α -amylase and
15 protease were applied.

16 There are several reports on the results from the microbiological assay being higher than
17 those from HPLC-methods [4–7]. However, not always a direct comparison with the same
18 samples was executed. As possible reasons for the differences observed in direct
19 comparisons either unidentified folates or substances being active in the microbiological
20 assay were hypothesized. [5,6]. In contrast to this, other studies did not find a difference
21 between these two methods [2]. To answer the question about a difference between the
22 microbiological assay and the LC-MS/MS, we performed a direct comparison. A further
23 objective was to investigate the need for the “trienzyme treatment” by surveying and if
24 necessary improving the efficiency of deconjugation. Moreover, we had a look on
25 additional peaks in the chromatograms of the MRM traces of 5-HCO-H₄folate and its

1 polyglutamates and investigated the reportedly different response of the microbiological
2 assay to the different vitamers[9–11].

3

4

5 **2. Experimental**

6 2.1. Synthesis of pyrazino-s-triazine (MeFox)

7 MeFox was synthesized according to Gapski et al. [12] with slight modifications. 1 mg 5-
8 CH₃-H₄folate was dissolved in 1 ml phosphate buffer (0.1 mol/l, pH 6 instead of 5), 50 µl
9 aqueous H₂O₂ (30%) was added and the solution was stirred for 2 h (remaining amount of
10 5-CH₃-H₄folate thereafter was 2 %). The residual H₂O₂ was removed by Na₂S₂O₃ (1 mol/l)
11 and a spatula of NaI. The absence of H₂O₂ was verified with a test strip. After purification
12 by semi-preparative HPLC (Pro C18 EC, 150 x 10 mm, 5 µm, 120 Å; gradient with 0.1 %
13 (v/v) formic acid and acetonitrile) the solution was lyophilized and the product dried over
14 concentrated sulphuric acid in a desiccator.

15

16 2.2. Folate analysis by LC-MS/MS

17 Distribution and amount of total folates were determined by LC-MS/MS according to a
18 previous publication [8]. Briefly, after addition of [²H₄]-labeled standards, the folates were
19 extracted in 10 ml MES-buffer (with 0.1 % (m/v) DTT and 0.1 % (v/v) mercaptoethanol as
20 a mixture or 1.3 % (v/v) mercaptoethanol or 0.1 % (m/v) DTT) and deconjugated by
21 addition of 150 µl rat serum and 2 ml chicken pancreas suspension (0.33 mg/2 ml) over
22 night. Afterwards 10 ml acetonitrile was added for precipitation of the proteins (mostly for
23 column care), then the sample was centrifuged and purified by SAX-SPE. The cleaned
24 extracts were measured with LC-MS/MS. To food samples, for which a higher amount of
25 enzymes for deconjugation was necessary, 300 µl rat serum and 2 ml of a more
26 concentrated chicken pancreas solution (2 mg/2 ml) were added, respectively 450 µl rat

1 serum and 4 ml of the chicken pancreas solution (2 mg/2ml). For investigating the
2 necessity of a di- or trienzyme treatment, 10 mg α -amylase per g food sample were added
3 and incubated for 2 h at 37°C. Thereafter, or alternatively, 3 mg protease per g food
4 sample were added and incubated for 4 h at 37°C. After the protease treatment the
5 enzymes were inactivated in a boiling water bath for 10 min, the sample extract was
6 cooled in an ice bath and the enzymes for deconjugation were added (see above). When
7 only the amylase treatment was applied no additional heat treatment was necessary.

8 In contrast to our publication mentioned above [8], a different HPLC column (Pro-C18, 250
9 x 3.2 mm, 3 μ m, 130 Å, YMC, Japan; precolumn: C18 4 x 2 mm, Phenomenex, Torrance,
10 USA), other mobile phases (A: 0.1% (v/v) formic acid, B: acetonitrile with 0.1 % formic
11 acid (v/v)) and an adapted gradient was used. The gradient elution started at 5 % B,
12 raising up to 10 % B in 5 minutes and staying at this composition for another 5 minutes,
13 followed by raising to 15 % B during 10 minutes and to 50 % in 2 minutes. This
14 composition was held for 4 minutes before being brought back to 5 % B in 2 minutes and
15 equilibrating the column for 9 minutes.

16 Additionally, MeFox, folate diglutamates and higher folate polyglutamates were included
17 into the LC-MS/MS-method. The specific MS parameters for the monoglutamates are the
18 same as described in [8], those for the diglutamates and higher polyglutamates are given
19 in table 1.

20 The response factors for quantifying the folate monoglutamates were the same as
21 described in [8], except for 10-HCO-H₂folate (1.149 ± 0.056 , internal standard: [²H₄]-10-
22 HCO-PteGlu) and 5,10-CH⁺-H₄folate (1.056 ± 0.057 , internal standard: [²H₄]-5-CH₃-
23 H₄folate). For the response factor of MeFox using [²H₄]-10-HCO-PteGlu as the internal
24 standard, seven concentration levels of MeFox along with a constant concentration of
25 [²H₄]-10-HCO-PteGlu were prepared in the elution buffer (5 % NaCl, 1 % ascorbic acid,
26 0.1 % DTT) with concentration ratios between 0.02:1 and 200:1.

1 For determining the response factors for the diglutamates, three different food stuffs
2 (broccoli, baguette and pea soup) were processed in two different ways. One approach
3 included the usual sample preparation with rat serum and chicken pancreas to generate
4 the monoglutamates and the second approach used only chicken pancreas to stop
5 deconjugation on the diglutamate level. The difference in the amount of both approaches
6 ($\Delta n_{\text{Monoglutamate}}$) represented the amount of the diglutamates in the second approach. The
7 response factors RF of the diglutamates to the corresponding [$^2\text{H}_4$]-labeled
8 monoglutamates (ISTD) were calculated as follows:

$$9 \quad RF = \frac{\text{Area}(\text{Diglutamate})}{\text{Area}(\text{ISTD})} \cdot \frac{\Delta n(\text{Monoglutamate})}{n(\text{ISTD})}$$

10 RFs are summarized in table 1.

11

12 2.3. Microbiological assay

13 In lab A the sample preparation was adapted to the LC-MS/MS method: 0.2 to 2 g food
14 sample were weighed into 50 ml PET centrifugal tubes and 10 ml MES-buffer was added.
15 The tubes were placed into a boiling water bath for 10 min and cooled in an ice bath
16 thereafter. Then, 150 μl rat serum and 2 ml chicken pancreas solution (0.33 mg/2 ml)
17 were added and the samples were incubated over night in a shaking water bath at 37°C.
18 The next day the samples were cooked in a boiling water bath for 10 min, the volume of
19 the samples was added to 20 ml and the tubes were centrifuged. The extract was diluted
20 with a phosphate buffer (0.05 mol/l, pH 7.2, with 0.1 g/l ascorbic acid and 1.3 % v/v
21 mercaptoethanol) and filtered with aseptic membrane filters (regenerated cellulose,
22 0.2 μm). For calibration, (6S)-5-CHO- H_4 folate or (6S)-5-CH $_3$ - H_4 folate was used. The
23 calibrants were dissolved in MES-buffer and diluted with the same phosphate buffer,
24 which was used for the food samples. 150 μl of the sample respectively 150 μl of the
25 standard solutions were pipetted into the cavities of the microbiological assay kit from R-

1 Biopharm (using *Lactobacillus rhamnosus*) and subsequently 150 µl of the assay medium
2 was added. The assay tray was sealed with a self-adhesive foil and the cavities were
3 incubated at 37°C in a small incubator. After 46 h the cavities were shaken overhead, the
4 foil was removed and the turbidity of the sample solutions was measured at 540 nm.

5 In lab B the same kit was used. Briefly, according to the description of the kit, a phosphate
6 buffer (0.05 mol/l, pH 7.2, 0.1 % ascorbate) was used for extraction and pig or chicken
7 pancreatin for deconjugation (2 h). Afterwards, the extracts were heated at 95°C for
8 30 minutes, centrifuged and diluted with sterile water. For calibration, (6S)-5-HCO-H₄folate
9 or PteGlu were used.

10 Lab C used a more classic version of the microbiological assay in test tubes, but with the
11 same microorganism (*Lactobacillus rhamnosus*). Briefly, chicken pancreas and phosphate
12 buffer A at pH 7.8 (0.1 mol/l, with 1% (m/v) sodium ascorbate: 13.8 g sodium dihydrogen
13 phosphate monohydrate, 10.0 g sodium ascorbate were diluted in 900 ml distilled water,
14 the pH was adjusted to 7.8 ± 0.1 with sodium hydroxide solution (5 % and 50 %) and the
15 volume was added to 1 L with distilled water) were added to the samples and incubated
16 approximately for 15 h at 37°C. The samples were centrifuged and diluted with another
17 phosphate buffer B (0.1 mol/l, with 0.1% (m/v) ascorbic acid: 6.96 g sodium hydrogen
18 phosphate dihydrate, 21.76 g sodium dihydrogen phosphate monohydrate and 2 g
19 ascorbic acid were diluted in distilled water and the volume was added to 2 L). Then the
20 medium of pH 6.1 (9.4 g Folic Acid Casei Medium from difco + 100 g phosphate buffer B)
21 was added in equal parts and the solution was heated for 7 min at 121°C. Afterwards, the
22 suspension of the microorganisms was added and the samples were incubated for
23 approximately 16 to 24h at 37°C. For calibration, (6S)-5-HCO-H₄folate and PteGlu were
24 used. Optical density was assessed at 546 nm.

25

1 3. Results

2 3.1. Optimization for the LC-MS/MS-method

3 3.1.1. Trienzyme treatment

4 The di- and trienzyme treatment is recommended for the extraction of folates from food
5 matrices. To verify the need for these additional treatments we added α -amalyse and/or
6 protease to the food extract of wheat germs, which are rich in starch and protein, before
7 adding the deconjugases (chicken pancreas plus rat serum). The results are presented in
8 figure 1.

9 The approach without an additional treatment, thus adding only the two deconjugases,
10 yielded the highest amount of total folate. This amount was not significantly different ($p >$
11 0.05) from the result of the second approach, where α -amylase was added. Besides,
12 adding protease gave a significant lower amount of total folates ($p < 0.01$) and adding
13 both enzymes (trienzyme treatment) an even lower amount ($p < 0.01$). This result has not
14 yet been optimized for deconjugation, which is described in the following paragraph.

15

16 3.1.2. Polyglutamates

17 To control the efficiency of deconjugation, the remaining amount of diglutamates was
18 quantified and additionally the sample solution was screened for the presence of higher
19 polyglutamates.

20 In various samples a complete deconjugation could be achieved, for example in corn
21 meal, orange juice and baguette bread ($14 \mu\text{g}/100\text{g}$, $25 \mu\text{g}/100\text{g}$ and $30 \mu\text{g}/100\text{g}$,
22 respectively) or at least a nearly complete deconjugation, for example in broccoli, spinach
23 and soybeans ($176 \mu\text{g}/100\text{g}$, $113 \mu\text{g}/100\text{g}$ and $245 \mu\text{g}/100\text{g}$). In contrast to this, in many
24 sample solutions of different foods di- and often even higher polyglutamates were still
25 present after deconjugation, for example in wheat germs and peas (figure 3).

1 Chromatograms of the polyglutamates of 5-CH₃-H₄folate in peas are shown in figure 2 (left
2 side).

3 In some cases the amount of diglutamates in relation to the sum of mono- and
4 diglutamates (Σ) was negligible, for example in broccoli with 0.3 % ($\Sigma = 176 \mu\text{g}/100\text{g}$),
5 basmati rice with 0.1 % ($\Sigma = 10 \mu\text{g}/100\text{g}$) and tomato juice with 0.4 % ($\Sigma = 19 \mu\text{g}/100\text{g}$), in
6 other cases the amount was within the range of analytical variation, e.g. cauliflower with
7 2 % ($\Sigma = 88 \mu\text{g}/100\text{g}$), lentils with 5 % ($\Sigma = 143 \mu\text{g}/100\text{g}$), potatoes with 4 % ($\Sigma =$
8 $18 \mu\text{g}/100\text{g}$), couscous with 2 % ($\Sigma = 17 \mu\text{g}/100\text{g}$), carrots with 3 % ($\Sigma = 55 \mu\text{g}/100\text{g}$) and
9 calf's liver with 2 % ($\Sigma = 663 \mu\text{g}/100\text{g}$). But in pistachios this quantification revealed up to
10 40 % ($\Sigma = 40 \mu\text{g}/100\text{g}$). In contrast to this, other foods revealed high amounts of
11 diglutamates such as mung beans with 29 % ($\Sigma = 422 \mu\text{g}/100\text{g}$), peanuts with 20 % ($\Sigma =$
12 $99 \mu\text{g}/100\text{g}$), camembert with 24 % ($\Sigma = 108 \mu\text{g}/100\text{g}$), wheat germs with 24 % ($\Sigma =$
13 $321 \mu\text{g}/100\text{g}$) and peas with 33 % ($\Sigma = 101 \mu\text{g}/100\text{g}$).

14 The screening for higher polyglutamates indicated that in the samples with a higher
15 amount of diglutamates often distinct amounts of tri- and higher polyglutamates were
16 present. Therefore, in these cases the total amount of folates was still higher than the sum
17 of the mono- and diglutamates. In the case of lentils, couscous and basmati rice there
18 were no other polyglutamates than diglutamates and, therefore, the sum of mono- and
19 diglutamates represented the total amount of folates. In the case of cauliflower, carrots
20 and calf's liver there was only a little residue of triglutamates, which can be neglected. But
21 there was a food sample (adzuki beans) which had only an amount of diglutamates of
22 4 %, which seemed not really high, but the screening of higher polyglutamates revealed a
23 noticeable amount of these.

24

25 3.1.3 Buffer optimization for deconjugation

1 Due to toxicity and unpleasant odor we wanted to replace mercaptoethanol (1.3 % v/v) by
2 DTT (0.1 % m/v), but the efficiency of deconjugation seemed worse (fig. 4). In fact, the
3 amount of diglutamates was not significantly different, but the amount of monoglutamates
4 appeared to be different ($p = 0.04$). When using a combined buffer of DTT and
5 mercaptoethanol (0.1 % each) a significant rise in the amount of monoglutamates ($p =$
6 0.002 based on DTT buffer and $p = 0,003$ based on MCE-buffer) and a significant
7 decrease in the amount of diglutamates ($p = 0.0014$ based on DTT-buffer and $p = 0.004$
8 based on MCE-buffer) could be detected. Nevertheless, the sum of mono- and
9 diglutamates was still the same.

10

11 3.1.4. Optimization of deconjugation

12 The effect of a smaller sample weight as well as a higher amount of deconjugases was
13 investigated to obtain a complete deconjugation. As can be seen in figure 3 a smaller
14 sample weight in combination with an additional amount of enzymes was effective. In
15 most cases 300 μl of rat serum instead of 150 μl plus a sixfold higher amount of chicken
16 pancreas was satisfactory. But in the case of the pistachios a very low sample weight of
17 0.12 g, 450 μl rat serum and twice the amount of the sixfold chicken pancreas solution
18 was necessary to reach an acceptable deconjugation (only 0.4% diglutamates were left,
19 but no polyglutamates). In conclusion, by adapting the sample weight and the amount of
20 enzymes for deconjugation, a successful deconjugation could be achieved.

21 Another possibility instead of rising the amount of enzymes is a longer incubation time.
22 For wheat germs a low sample weight and incubation over two nights gave the same
23 result as the addition of the higher amount of enzymes.

24

25 3.1.4 Pyrazino-s-triazine (MeFox)

1 In several food stuffs an additional peak was visible in the MRM chromatogram of the 5-
2 CHO-H₄folate trace (MRM 474 → 327). Interestingly, in each MRM chromatogram of the
3 respective polyglutamates of 5-CHO-H₄folate (fragments always after splitting the whole
4 glutamate chain) an additional peak was visible, too. The same peak in the 5-CHO-
5 H₄folate trace appeared after incubating 5-CH₃-H₄folate in the context of investigating the
6 stability of 5-CH₃-folate during digestion in a simulation model [13], i.e. at pH 2 in diluted
7 hydrochloric acid for 2 h followed by adding sodium hydrogen carbonate, at pH 5.5 (for
8 3 h) and 6.5 (an additional hour). When adding ascorbic acid, the amount of this
9 substance decreased. Therefore, it was assumed, that this substance, with the MRM
10 transition of 474 → 327 is an oxidation product of 5-CH₃-H₄folate. An oxidation product of
11 5-CH₃-H₄folate has already been reported, i. e. a substance with a pyrazino-s-triazine
12 structure [14], which was formerly misidentified as 4a-hydroxy-5-methyl tetrahydrofolate
13 (hmTHF) [12,15]. According to Hannisdal et al. [16] and Fazili et al. [17] this oxidation
14 product, which has also been abbreviated MeFox, shows the MRM-transition of 474 →
15 327. When looking at the fragments of the MS/MS-spectrum of our peak, we found it in
16 agreement with the structure of this pyrazino-s-triazine. Therefore, we synthesized MeFox
17 and compared it with the substance found in several food stuffs by its retention time, UV-
18 spectrum and MS/MS-spectrum. All were in good agreement. Additionally, MeFox is
19 known to show no activity in the microbiological assay [18], which we could also confirm.
20 To get an impression about its amount in foods, we quantified MeFox using [²H₄]-10-CHO-
21 PteGlu as the internal standard. We chose this standard due to the fact that it shows no
22 interconversions during analysis and already showed good validation results in the
23 analysis of 10-CHO-H₂folate [8], which has a similar retention time to MeFox. The results
24 are shown in table 2. MeFox can be found in different kinds of food stuffs: in fresh
25 vegetables like cauliflower, carrots and peas; in dried legumes like adzuki beans and soy
26 beans; in grain respectively grain products like wheat germs, couscous, oat flakes and

1 spaghetti and in nuts and similar food stuffs like hazelnut , almonds, walnuts and peanuts.
2 In some samples the amount of MeFox was quite high, for example in wheat germs, oat
3 flakes and walnuts and the ratio to the amount of total folate was highest in couscous, rice
4 and spaghetti. This was not surprising as these products are low in ascorbic acid and
5 were dried or further processed.

6

7 3.2. Microbiological assay

8 As mentioned in the introduction the response of the microbiological assay to food folate
9 vitamers is different [9–11]. To investigate this effect in more detail, we compared the
10 results of folate quantitations in three different labs using different folates for calibration.
11 As can be seen in figure 5, in all three labs there were distinct differences with respect to
12 the folate utilized for calibration. The results for the total amount of folates differed
13 depending on the folate and the lab between 7 and 37 %.

14

15 3.3. Comparison between LC-MS/MS and microbiological assay

16 After optimizing the LC-MS/MS method particularly with regard to a complete
17 deconjugation, several food stuffs were analyzed both with LC-MS/MS and with the
18 microbiological assay. For calibrating the microbiological assay both PteGlu and 5-CHO-
19 H₄folate were used (lab C), except from broccoli, for which we used 5-CH₃-H₄folate as a
20 calibrant instead of PteGlu (lab A). The comparison is shown in table 3 and the folate
21 composition of the corresponding food in table 4.

22 For broccoli, the vegetable mix (BCR 485), wheat germs, chick peas and pistachios there
23 was no significant difference between the two methods when using the folate with the
24 highest proportion for calibration. Also for spaghetti and mung beans there was no
25 significant difference between the results of the LC-MS/MS measurement and the

1 microbiological assay when using a folate as a calibrant, which was not the one with the
2 highest proportion.

3

4

5 **4. Discussion**

6 As many publications report a smaller amount of folates analyzed by HPLC-methods
7 compared to the microbiological assay we optimized our sample preparation for LC-
8 MS/MS for completeness of extraction and deconjugation before executing the
9 comparison.

10 An often recommended method is the di- or trienzyme treatment. When we compared our
11 usual treatment before LC-MS/MS including only deconjugases (“single treatment”) with
12 the additional treatments with α -amylase and / or protease (“dienzyme” and “trienzyme
13 treatment”) the latter two showed no benefits compared to the “single treatment”. We
14 assume that the application of chicken pancreas for deconjugation made an additional
15 enzyme treatment with α -amylase and / or protease obsolete due to the fact that this
16 source of deconjugase also contains further enzymes like amylases [19].

17 Then we had a look on the completeness of folates’ deconjugation by analyzing the
18 amount of diglutamates and screening for the presence of higher polyglutamates by LC-
19 MS/MS. In several samples the deconjugation was complete, whereas in other samples
20 the deconjugation activity was not sufficient (diglutamates were present and in some
21 samples tri- and higher polyglutamates, too). However, the amount was not expected to
22 be too high, for example in pistachios, buck wheat and camembert. One reason could be
23 the presence of other polyglutamates, which were deconjugated by the same enzymes,
24 for example MeFox, the oxidation product of 5-CH₃-H₄folate. However, in camembert
25 neither the amount of folates nor the amount of MeFox should be too high. Therefore,
26 deconjugation inhibitors may be present, as was already mentioned in earlier publications

1 [7,20]. As such, glutamic acid [21], citric acid [22] or even fatty acids could be conceivable.
2 The latter can inhibit the activity of enzymes [23], so that an inhibition of the deconjugation
3 enzymes in rat serum and chicken pancreatin is likely. Even the components of the buffer
4 can influence the deconjugation, as revealed by our experiment with different amounts of
5 DTT and mercaptoethanol. It seemed, that mercaptoethanol can both promote and in
6 higher amounts inhibit the deconjugation. The promoting effect of mercaptoethanol on the
7 deconjugases of rat serum was already mentioned in [24].

8 In most cases, analyzing the amount of diglutamates gave a good hint about the efficiency
9 of deconjugation. Sometimes already the sum of mono- and diglutamates was sufficient
10 for calculating the amount of total folates, especially when the amount of diglutamates
11 was low and no higher polyglutamates were present. But there are a few exceptions, for
12 example adzuki beans. Here, the amount of diglutamates was low, but the screening for
13 higher polyglutamates and an analysis with a higher amount of deconjugases revealed
14 that the efficiency of deconjugation was lower than expected by the amount of
15 diglutamates. Therefore, this food stuff showed that a screening for higher polyglutamates
16 can sometimes be essential. Probably, in these cases the deconjugase of the chicken
17 pancreas is more inhibited than the deconjugase of the rat serum.

18 In those foods with higher amounts of diglutamates, the sum of mono- and diglutamates
19 was not enough to represent the total amount of folates. Therefore, the method needed to
20 be optimized. The reduction of the samples size can improve the efficiency of
21 deconjugation. But to keep the sample amount in several samples feasible, a higher
22 addition of deconjugation enzymes was necessary. Prolonging the incubation time is
23 another possibility to improve deconjugation. With respect to the stability of folates and the
24 total analysis time, this alternative is recommended for food stuffs with low amounts,
25 where reducing the sample amount and subtraction of the even higher amount of the

1 intrinsic folates of the added enzymes would result in a higher uncertainty of
2 measurement.

3 We could confirm the additional peaks both in the chromatograms of the 5-CHO-H₄folate
4 MRM trace and the MRM traces of the corresponding polyglutamates to be the oxidation
5 product of the folate 5-CH₃-H₄folate, namely MeFox. But MeFox has no microbiological
6 activity and, therefore, cannot be a reason for the differences between HPLC-methods
7 and the microbiological assay. Nevertheless, this substance seemed to be related to the
8 stability of 5-CH₃-H₄folate in blood during storage [25, 26] and also to the stability of the
9 latter folate in food. In many samples the amount of MeFox is considerably higher than the
10 amount of folates, especially when they have been processed and are low in ascorbic
11 acid. Nevertheless, even in fresh carrots, the amount respectively the ratio to folates is not
12 negligible. It could be interesting from the technological and nutritive point of view to have
13 a closer look on the formation of MeFox with respect to its reduction or the question of its
14 formation during plant growth.

15 When regarding the microbiological assay, we confirmed in three labs that there were
16 differences in the response to the single folate vitamers. One reason could be the pH
17 during incubation. Lab A and B used a commercial kit from the same manufacturer. Here
18 the buffer capacity was probably not high enough, due to the dropping of pH from 6.5 to
19 5.2 with rising amount of folates. In contrast to this, in lab C a buffered pH of 6.1 ± 0.1 was
20 used. That pH is similar to the recommended pH of 6.2 by Phillips and Wright [9], at which
21 the difference in response shall be virtually minimal. Nevertheless, distinct differences
22 between the results based on calibration with 5-CHO-H₄folate and PteGlu were observed.
23 The stability of the used folates cannot be the reason, because 5-CHO-H₄folate and
24 PteGlu are relatively stable folates. Moreover, in lab A using 5-CH₃-H₄folate for calibration,
25 a folate being more prone to oxidative degradation, mercaptoethanol was used to prevent
26 oxidation thus excluding lack of stability as a reason for the differences in the response.

1 Due to the fact that even when holding the pH range very close to the recommended
2 value, the response of the microbiological assay differs between the folate vitamers, it was
3 advantageous to use the folate with the highest proportion for calibration (“main” folate).
4 The comparison showed that the results of LC-MS/MS are even higher on average than
5 the results from the microbiological assay calibrated with PteGlu, which is the calibrant
6 with the highest utilization. This is in contrast to the presumption that the microbiological
7 assay often results in significantly higher amounts. For broccoli a really good correlation
8 between the microbiological assay and the SIDA was achieved when using 5-CH₃H₄folate
9 for calibration. With an amount of 81%, 5-CH₃-H₄folate can be considered as the “main”
10 folate. Calibrating with 5-CHO-H₄folate would result in an overestimation. As well as in
11 broccoli, the “main” folate in the vegetable mix is 5-CH₃-H₄folate. In this case PteGlu was
12 used for calibration as this is also the usual calibrant in the reference microbiological
13 assay. This vegetable mix is a certified reference material (BCR 485) with a certified value
14 for total folate of $315 \pm 28 \mu\text{g}/100\text{g}$. Due to the fact that both, the result from LC-MS/MS
15 and the microbiological assay via PteGlu, showed no significant difference, we assume
16 that the responses of the used microbiological assay for PteGlu and 5-CH₃-H₄folate were
17 similar. Therefore, PteGlu could be used instead of 5-CH₃-H₄folate. In these two cases, an
18 explicit “main”-folate existed, which yielded results showing no significant difference from
19 the results by LC-MS/MS.

20 For wheat germs, chick peas and pistachios the folate with the highest proportion was 5-
21 CHO-H₄folate. Using this folate as a calibrant gave results also being not significantly
22 different from the results by LC-MS/MS. A calibration with the most frequently used
23 PteGlu resulted in an underestimation. Therefore, all the mentioned food stuffs confirm
24 that it is necessary to calibrate the microbiological assay with 5-HCO-H₄folate, if it is the
25 vitamer with the highest proportion, otherwise with PteGlu or 5-CH₃-H₄folate.
26 Nevertheless, earlier publications already used 5-HCO-H₄folate as a calibrant [27]. In the

1 respective samples, 5-HCO-H₄folate had the highest proportion, but only with an average
2 of 35 %, so that the other folates with an amount of 65 % could have influenced the
3 overall response in the microbiological assay. Therefore, we assume that having no
4 explicit “main” folate could over- or underestimate the results. This most likely happened
5 to our result for spaghetti. Here the microbiological assay using 5-HCO-H₄folate as a
6 calibrant produced an amount for total folate distinctively higher than the LC-MS/MS,
7 whereas using PteGlu showed no significant difference. The distribution in spaghetti
8 revealed a similar amount of 5-HCO-H₄folate and 10-HCO-PteGlu. The latter shows a
9 microbiological activity of 95 % compared to PteGlu, which is not significantly different.
10 PteGlu and 10-HCO-PteGlu represent approximately half of the folate vitamers.
11 Considering that 5-CH₃-H₄folate has a similar response as PteGlu, as evident from the
12 results of the vegetable mix mentioned above, more than 50 % of the vitamers would be
13 better represented by the calibration using PteGlu.

14 For mung beans the calibration with 5-CHO-H₄folate and with PteGlu gave lower results
15 compared to the LC-MS/MS, although the result via 5-CHO-H₄folate was not significantly
16 different. But actually 5-CHO-H₄folate was not the “main” folate. 5-CH₃-H₄folate was the
17 largest fraction amounting to 45 %. Therefore, a calibration with PteGlu should be
18 sufficient. But the proportion of 5-CHO-H₄folate is 30 % and additionally H₄folate is
19 accounting for 14 %. We assume that the latter folate is degraded to a certain amount
20 during incubation, at least in lab B and C (in lab A mercaptoethanol was used for
21 stabilization). Besides, the results for the mung beans can be explained when taking into
22 account that the deconjugation to the diglutamates was not complete in combination with
23 the knowledge, that according to Tamura et al. [1] the response of the microbiological
24 assay to higher polyglutamates is lower than for mono-, di- and triglutamates. Even for
25 wheat germs, chick peas and pistachios an incomplete deconjugation as well as a certain
26 loss of H₄folate could have lowered the results of the microbiological assay. Nevertheless,

1 this does not disagree with the former observation that LC-MS/MS and the microbiological
2 assay achieve results being not significantly different, because using 5-HCO-H₄folate
3 solely as a calibrant could have very probably overestimated the amount and
4 compensated for the converse effects. Therefore, in some cases it may be helpful to use a
5 mixture of calibrants with a folate composition representing the distribution of the food.
6 Nevertheless, we recommend to use a calibrant representing the main vitamer for the
7 microbiological assay and a sufficient deconjugation for both methods. Then this will give
8 similar results of the microbiological assay and the LC-MS/MS coming close to the “true”
9 value. However, as the microbiological assay is not able to reveal the vitamer distribution
10 and the correct calibrant, we see significant drawbacks for the method in those samples.

11

12 **5. Conclusion**

13 From our results, several reasons for the differences between the microbiological assay
14 and LC-MS/MS can be concluded.

15 First, insufficient deconjugation can lead to lower results in HPLC methods. As an
16 incomplete deconjugation hardly can be predicted, a screening for polyglutamates or their
17 respective quantitation is recommended.

18 Second, the microbiological assay may give inaccurate results due to different responses
19 and stabilities of the single vitamers. This problem hardly can be solved with the
20 microbiological assay alone, information about vitamer distribution from other methods are
21 necessary.

22 Third, and what we did not mentioned up to now, HPLC methods can be inaccurate when
23 using unequivocal UV extinction coefficients for calibration of the single vitamers.
24 Significant disagreement of this values appears from several reports [28,29] and we
25 recommend to use the values from the most recent study [29].

26

1 **6. Acknowledgments**

2 We thank r-biopharm i.a. for supplying the microbiological assay kits we used in our lab
3 and the lab “Institut Kirchhoff Berlin” for their intense analysis with the microbiological
4 assay.

7 **7. References**

- 8 [1] T. Tamura, M.A. Williams, Y.S. Shin, E.L.R. Stokstad, Lactobacillus casei Response to
9 Pteroylpolyglutamates, Analytical Biochemistry 49 (1972) 517–521.
- 10 [2] C.M. Pfeiffer, L.M. Rogers, J.F. Gregory, Determination of folate in cereal-grain food
11 products using trienzyme extraction and combined affinity and reversed-phase liquid
12 chromatography, J. Agric. Food Chem. 45 (1997) 407–413.
- 13 [3] J.I. Martin, W.O. Landen, A.G. Soliman, R.R. Eitenmiller, Application of a Tri-Enzyme
14 Extraction for Total Folate Determination in Foods, J. Assoc. Off. Anal. Chem. 73
15 (1990) 805–808.
- 16 [4] T.H. Hyun, T. Tamura, Trienzyme Extraction in Combination with Microbiologic Assay
17 in Food Folate Analysis: An Updated Review, Experimental Biology and Medicine 230
18 (2005) 444–454.
- 19 [5] S. Ruggeri, L.T. Vahteristo, A. Aguzzi, P. Finglas, E. Carnovale, Determination of folate
20 vitamers in food and in Italian reference diet by high-performance liquid
21 chromatography, Journal of Chromatography A 855 (1999) 237–245.
- 22 [6] M.S. Kariluoto, L.T. Vahteristo, V.I. Piironen, Applicability of microbiological assay and
23 affinity chromatography purification followed by high-performance liquid
24 chromatography (HPLC) in studying folate contents in rye, Journal of the Science of
25 Food and Agriculture 81 (2001) 938–942.

- 1 [7] M. Rychlik, K. Englert, S. Kapfer, E. Kirchhoff, Folate contents of legumes determined
2 by optimized enzyme treatment and stable isotope dilution assays, *Journal of Food*
3 *Composition and Analysis* 20 (2007) 411–419.
- 4 [8] C. Ringling, M. Rychlik, Analysis of seven folates in food by LC-MS/MS to improve
5 accuracy of total folate data, *European Food Research and Technology* 236 (2013)
6 17–28.
- 7 [9] D.R. Phillips, A.J. Wright, Studies on the response of *Lactobacillus casei* to different
8 folate monoglutamates, *British Journal of Nutrition* 47 (1982) 183–189.
- 9 [10] C.M. Pfeiffer, Z. Fazili, L. McCoy, M. Zhang, E.W. Gunter, Determination of folate
10 vitamers in human serum by stable-isotope-dilution tandem mass spectrometry and
11 comparison with radioassay and microbiologic assay, *Clinical Chemistry* 50 (2004)
12 423–432.
- 13 [11] W. Weber, S. Mönch, M. Rychlik, S. Stengl, Quantitation of Vitamins using
14 Microbiological Assays in Microtiter Formats, in: M. Rychlik (Ed.) *Fortified foods with*
15 *vitamins. Analytical concepts to assure better and safer products*, Wiley-VCH,
16 Weinheim, 2011 pp. 37–64.
- 17 [12] G.R. Gapski, J.M. Whiteley, F.M. Huennekens, Hydroxylated Derivatives of 5-Methyl-
18 5,6,7,8-tetrahydrofolate, *Biochemistry* 10 (1971) 2930–2934.
- 19 [13] C. Ringling, M. Rychlik, Simulation of food folate digestion and bioavailability of an
20 oxidation product of 5-Methyltetrahydrofolate, in progress.
- 21 [14] J.A. Jongejan, H.I.X. Mager, W. Berends, Autoxidation of 5-Alkyl-
22 Tetrahydropteridines. The Oxidation Product of 5-Methyl-THF, in: R.L. Kisliuk, G.M.
23 Brown (Eds.) *Chemistry and Biology of Pteridines*, Elsevier/North-Holland, New York,
24 1979 pp. 241–246.
- 25 [15] J.M. Whiteley, A. Russell, Structural reassignment of the peroxide oxidation-product
26 of 5-methyl-5,6,7,8-tetrahydrofolate, *Biochem Bioph Res Co* 101 (1981) 1259–1265.

- 1 [16] R. Hannisdal, P.M. Ueland, A. Svardal, Liquid Chromatography-Tandem Mass
2 Spectrometry Analysis of Folate and Folate Catabolites in Human Serum, *Clinical*
3 *Chemistry* 55 (2009) 1147–1154.
- 4 [17] Z. Fazili, C.M. Pfeiffer, Accounting for an Isobaric Interference Allows Correct
5 Determination of Folate Vitamers in Serum by Isotope Dilution-Liquid Chromatography-
6 Tandem MS, *Journal of Nutrition* 143 (2013) 108–113.
- 7 [18] K.R. Ratanasthien, J.A. Blair, R.J. Leeming, W.T. Cooke, V. Melikian, Serum folates
8 in man, *Journal of Clinical Pathology* 30 (1977) 438–448.
- 9 [19] J.C. Pedersen, Comparison of γ -glutamyl hydrolase (conjugase; EC 3.4.22.12) and
10 amylase treatment procedures in the microbiological assay for food folates, *British*
11 *Journal of Nutrition* 59 (1988) 261–271.
- 12 [20] R. Engelhardt, J.F. Gregory, Adequacy of Enzymatic Deconjugation in Quantification
13 of Folate in Foods, *J. Agric. Food Chem.* 38 (1990) 154–158.
- 14 [21] A. Kazenko, M. Laskowski, On the specificity of chicken pancreas conjugase
15 (gamma-glutamic acid carboxypeptidase), *Journal of Biological Chemistry* 173 (1948)
16 217–221.
- 17 [22] D.M. Goli, J.T. Vanderslice, Investigation of the Conjugase Treatment Procedure in
18 the Microbiological Assay of Folate, *Food Chemistry* 43 (1992) 57–64.
- 19 [23] S.V. Pande, J.F. Mead, Inhibition of Enzyme Activities by Free Fatty Acids, *Journal of*
20 *Biological Chemistry* 243 (1968) 6180–6185.
- 21 [24] D.W. Horne, C.L. Krumdieck, C. Wagner, Properties of Folic-Acid γ -Glutamyl-
22 Transferase Hydrolase (Conjugase) in Rat Bile and Plasma, *Journal of Nutrition* 111
23 (1981) 442–449.
- 24 [25] R. Hannisdal, P.M. Ueland, S.J.P.M. Eussen, A. Svardal, S. Hustad, Analytical
25 Recovery of Folate Degradation Products Formed in Human Serum and Plasma at
26 Room Temperature, *Journal of Nutrition* 139 (2009) 1415–1418.

- 1 [26] Z. Fazili, J.R.D. Whitehead, N. Paladugula, C.M. Pfeiffer, A high-throughput LC-
2 MS/MS method suitable for population biomonitoring measures five serum folate
3 vitamers and one oxidation product, *Analytical and Bioanalytical Chemistry* 405 (2013)
4 4549–4560.
- 5 [27] S. Kariluoto, M. Edelmann, V. Piironen, Effects of Environment and Genotype on
6 Folate Contents in Wheat in the HEALTHGRAIN Diversity Screen, *J. Agric. Food*
7 *Chem.* 58 (2010) 9324–9331.
- 8 [28] R.L. Blakley (Ed.), *The Biochemistry of the Folic Acid and Related Pteridines*, North-
9 Holland Publishing Company, Amsterdam, 1969.
- 10 [29] S. Moench, M. Rychlik, Improved Folate Extraction and Tracing Deconjugation
11 Efficiency by Dual Label Isotope Dilution Assays in Foods, *J. Agric. Food Chem.* 60
12 (2012) 1363–1372.
- 13 [30] S.W. Souci, W. Fachmann, H. Kraut, E. Kirchoff, *Food composition and nutrition*
14 *tables. Die Zusammensetzung der Lebensmittel Nährwert-Tabellen = La composition*
15 *des aliments. Tableaux des valeurs nutritives*, Medpharm Scientific Publishers,
16 Stuttgart, 2008.
- 17 [31] Deutsches Ernährungsberatungs- und -informationsnetz: DEBInet - ernaehrung.de,
18 Bundeslebensmittelschlüssel (BLS II.3 und BLS 3.01).
19 <http://www.ernaehrung.de/lebensmittel/> (01.07.2014).
- 20

1 **Tables**

2 Table 1: MRM transitions and corresponding parameters for MeFox, folate diglutamates
 3 and higher polyglutamates. As internal standards, the corresponding [²H₄]-labeled folate
 4 monoglutamates were used. However, for MeFox and 10-HCO-H₂PteGlu₂ [²H₄]-labeled
 5 10-HCO-PteGlu served as internal standard.

Compound	Precursor ion (m/z)	Product ion ¹ (m/z)	DP ²	EP ³	CE ⁴	CXP ⁵	RF ± s ⁶
MeFox (pyrazino-s-triazine)	474	327	66	10	29	14	1.496 ± 0.063
5-CH ₃ -H ₄ PteGlu ₂	589	313	60	10	40	8	0.721 ± 0.071
		180	60	10	65	10	
5-CH ₃ -H ₄ PteGlu ₃	718	313	70	10	50	8	
		180	70	10	75	10	
5-CH ₃ -H ₄ PteGlu ₄	847	313	80	10	60	8	
		180	80	10	85	10	
5-CH ₃ -H ₄ PteGlu ₅	976	313	90	10	70	8	
		180		10	95	10	
5-CH ₃ -H ₄ PteGlu ₆	1105	313	90	10	80	8	
5-CH ₃ -H ₄ PteGlu ₇	1234	313	90	10	90	8	
5-HCO-H ₄ PteGlu ₂	603	327	61	10	35	14	0.583 ± 0.052
		299	61	10	57	16	
5-HCO-H ₄ PteGlu ₃	732	327	80	10	50	14	
		299	80	10	60	16	
5-HCO-H ₄ PteGlu ₄	861	327	80	10	60	14	
		299	80	10	70	16	
5-HCO-H ₄ PteGlu ₅	990	327	70	10	70	14	
		299	70	10	80	16	
5-HCO-H ₄ PteGlu ₆	1119	327	70	10	80	14	
5-HCO-H ₄ PteGlu ₇	1248	327	70	10	90	14	
10-HCO-PteGlu ₂	599	295	70	15	50	16	0.774 ± 0.131
	599	176	70	15	70	8	
10-HCO-PteGlu ₃	728	295	70	15	55	16	
	728	176	70	15	80	8	
10-HCO-PteGlu ₄	857	295	70	15	65	16	
	857	176	70	15	90	8	
10-HCO-PteGlu ₅	986	295	70	15	75	16	
	986	176	70	15	100	8	
H ₄ PteGlu ₂	575	299	60	10	35	16	0.772 ± 0.026
	575	166	60	10	65	8	
H ₄ PteGlu ₃	704	299	60	10	50	16	
	704	166	60	10	80	8	

H ₄ PteGlu ₄	833	299	70	10	60	16		
	833	166	70	10	90	8		
H ₄ PteGlu ₅	962	299	80	10	70	16		
	962	166	80	10	100	8		
PteGlu ₂	571	295	60	8	30	14	0.480 ±	0.068
		176	60	8	65	8		
PteGlu ₃	700	295	60	8	40	14		
		176	60	8	70	8		
PteGlu ₄	829	295	70	8	50	14		
		176	70	8	80	8		
PteGlu ₅	958	295	80	8	60	14		
		176	80	8	90	8		
10-HCO-H ₂ PteGlu ₂	601	178	75	10	45	8	0.183 ±	0.056
		297	75	10	40	16		

1 ¹ Dwell time: 50 s, ² DP: Declustering potential (in V), ³ EP: Entrance Potential (in V), ⁴ CE: Collision Energy
2 (in V), ⁵ CXP: Cell Exit Potential (in V), ⁶ RF: response factor ± standard deviation

3

4

5

6

7 Table 2: Amount of MeFox and ascorbic acid in different foods. Additionally, the ratio of
8 MeFox to total folate is presented.

foodstuff	MeFox (in µg/100 g)	Ratio of MeFox to total folate	ascorbic acid (in mg/100 g)
carrot juice	159	8.6	3.8 ¹
carrots	364	6.2	7 ¹
cauliflower	19	0.19	64 ¹
peas	113	0.96	25 ¹
soy beans	187	0,57	34 ²
adzuki beans	551	2.5	
whet germs	2581	5.4	0 ²
buckwheat	140	2.4	0 ¹
couscous	591	34	0 ²
corn semolina	139	9.0	0 ¹

pearl barley	54	1.7	0 ¹
oatmeal	1182	30	0 ¹
millet	216	5.1	0 ²
basmati rice	759	59	0 ¹
spaghetti	767	41	0 ¹
potatoes	13	0.61	17 ¹
walnuts	2425	17	2.6 ¹
hazelnuts	597	8.4	3 ¹
peanuts	1394	8.6	0 ¹
almonds	95	3.1	3.7 ¹
pistachios	943	7.3	7 ¹

1 ¹ Souci, Fachmann, Kraut [30], ² Bundeslebensmittelschlüssel [31]

2

3 Table 3: Amount of total folate (calculated as PteGlu in µg/100g) in different kinds of food
4 analyzed by two methods: LC-MS/MS and microbiological assay (with three different
5 folates for calibration; results from lab C, except the result for broccoli with 5-CH₃-H₄folate
6 as calibrant, which was from lab A). The results from the microbiological assay in bold
7 characters are not significantly different (p>0.05) from the results of LC-MS/MS.

food stuff	LC-MS/MS	Microbiological assay		
		5-CH ₃ -H ₄ folate	5-CHO-H ₄ folate	PteGlu
broccoli	183	188	250	-
vegetable mix	336	-	459	311
wheat germs	392	-	438	277
chick peas	393	-	362	254
pistachios	112	-	124	93
spaghetti	17	-	28	20
mung beans	433	-	367	238

8

9

1 Table 4: Folate composition in different kinds of food (in %).

food stuff	5-CH ₃ - H ₄ folate	5-CHO- H ₄ folate	10- CHO- PteGlu	H ₄ folate	PteGlu	10- HCO- H ₂ folate	5,10- CH ⁺ - H ₄ folate
broccoli	81	13	3	2	0	1	1
vegetable mix	96	1	0	2	1	0	0
wheat germs	9	68	9	1	12	1	1
chick peas	28	49	6	11	4	1	2
pistachios	22	56	5	10	4	2	1
spaghetti	5	42	40	1	9	1	1
Mung beans	45	30	5	14	2	3	2

2

1 **Legends to the figures**

2

3 Figure 1: Amount of folates (Sum of the single vitamers, monoglutamates; calculated as
4 PteGlu in $\mu\text{g}/100\text{g}$) after different sample preparations or after enzyme
5 treatments in addition to the deconjugation step. This treatment has not yet
6 been optimized for deconjugation.

7

8 Figure 2: Chromatograms of $5\text{-CH}_3\text{-H}_4\text{PteGlu}_{1-5}$ in a sample solution of peas before (left
9 side) and after optimization of deconjugation (right side).

10

11 Figure 3: Amount of mono- and diglutamates (calculated as PteGlu) of different types of
12 food after original sample preparation and with varying sample weights and an
13 extra addition of deconjugases (+ E).

14

15 Figure 4: Comparison of the amount of folates in wheat germs calculated from mono-
16 and diglutamates (calculated as PteGlu) after the deconjugation step using
17 different thiols or a combination of these in the extraction buffer.

18

19 Figure 5: Results from the microbiological assay via calibration with different vitamers in
20 three different labs (A, B and C).

21

22

23

24

25

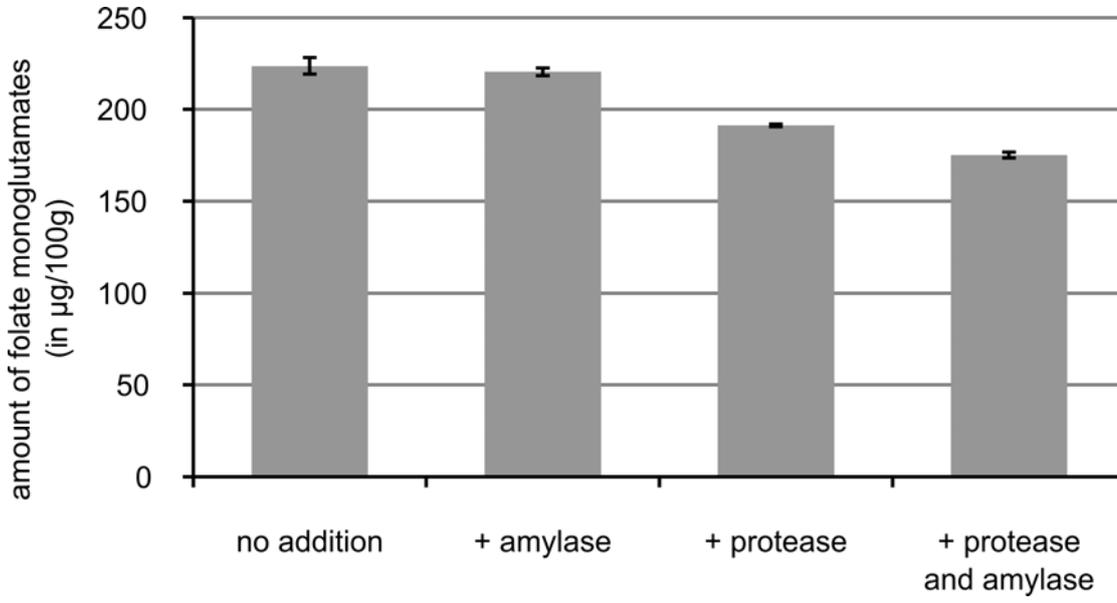
26

1 **Figures**

2

3 Figure 1

4



5

6

7

8

9

10

11

12

13

14

15

16

17

18

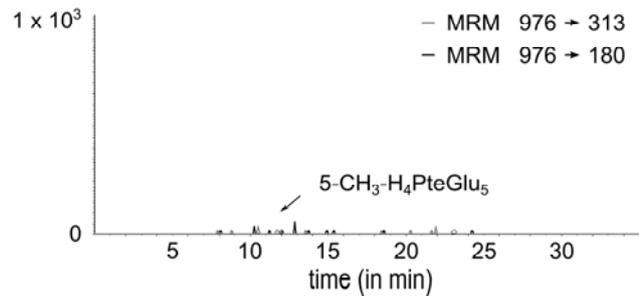
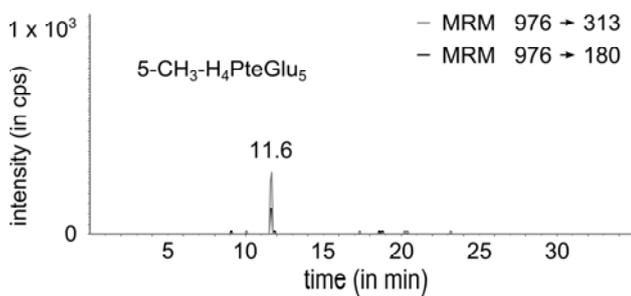
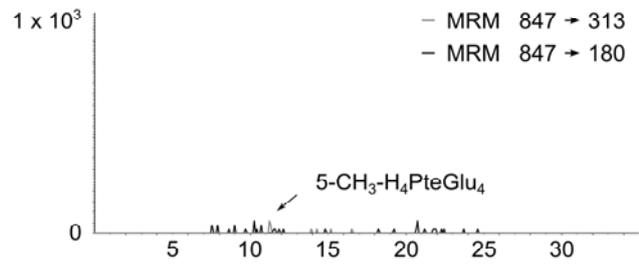
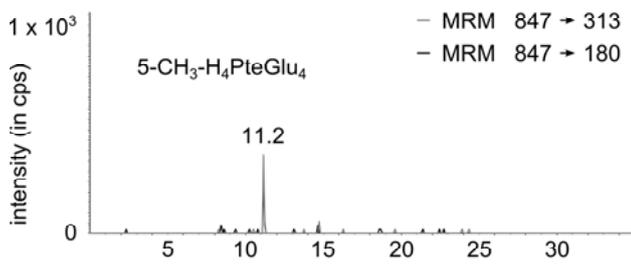
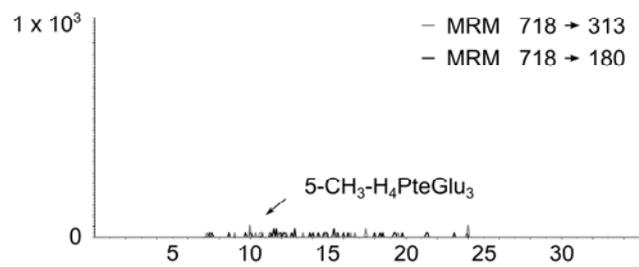
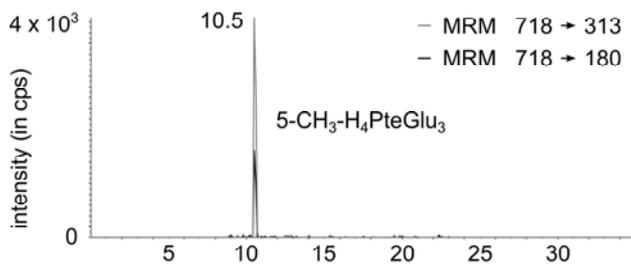
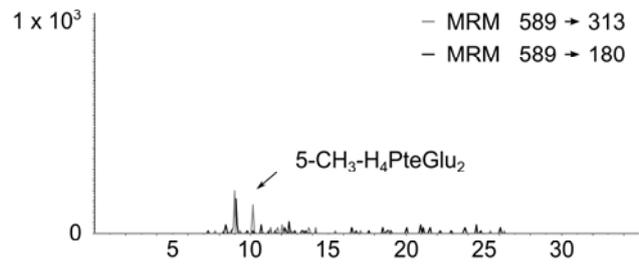
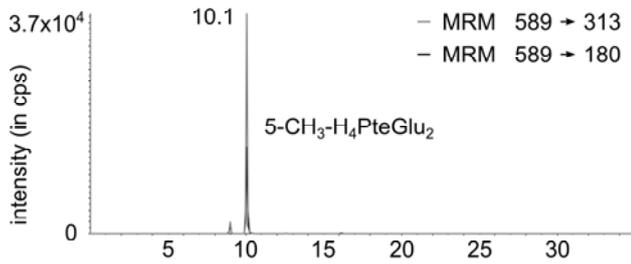
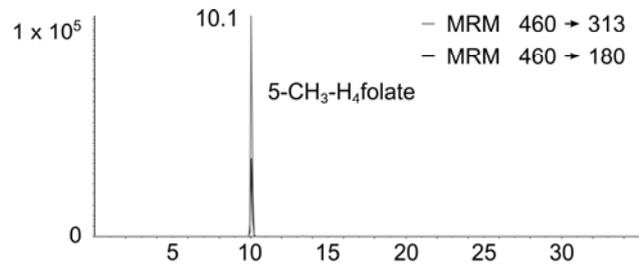
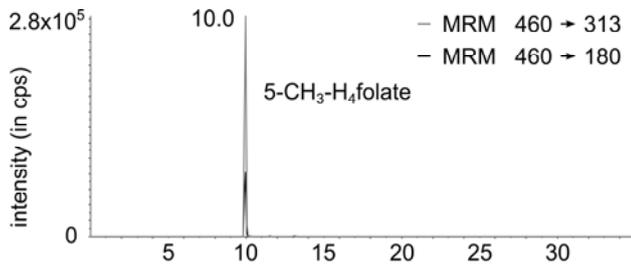
19

20

21

1 Figure 2

2



3

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16

Figure 3

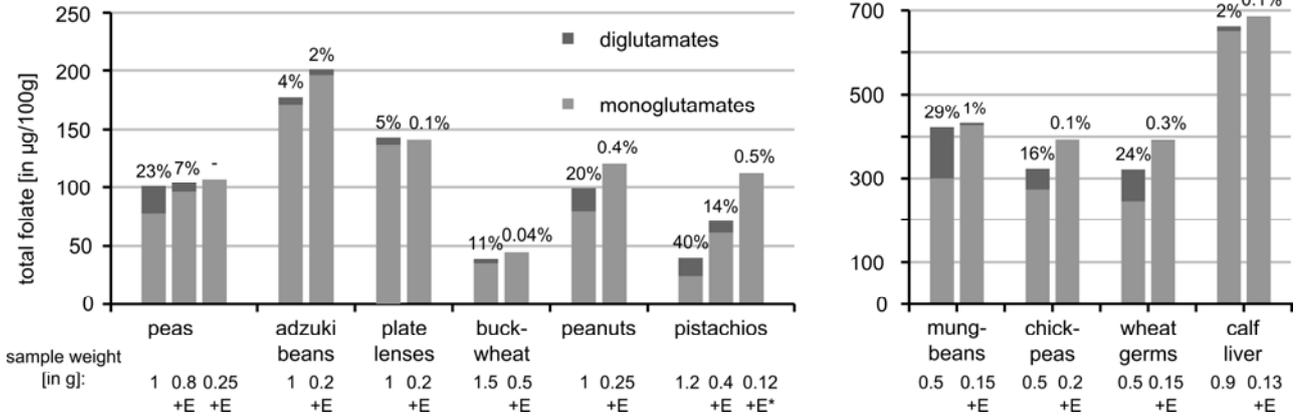
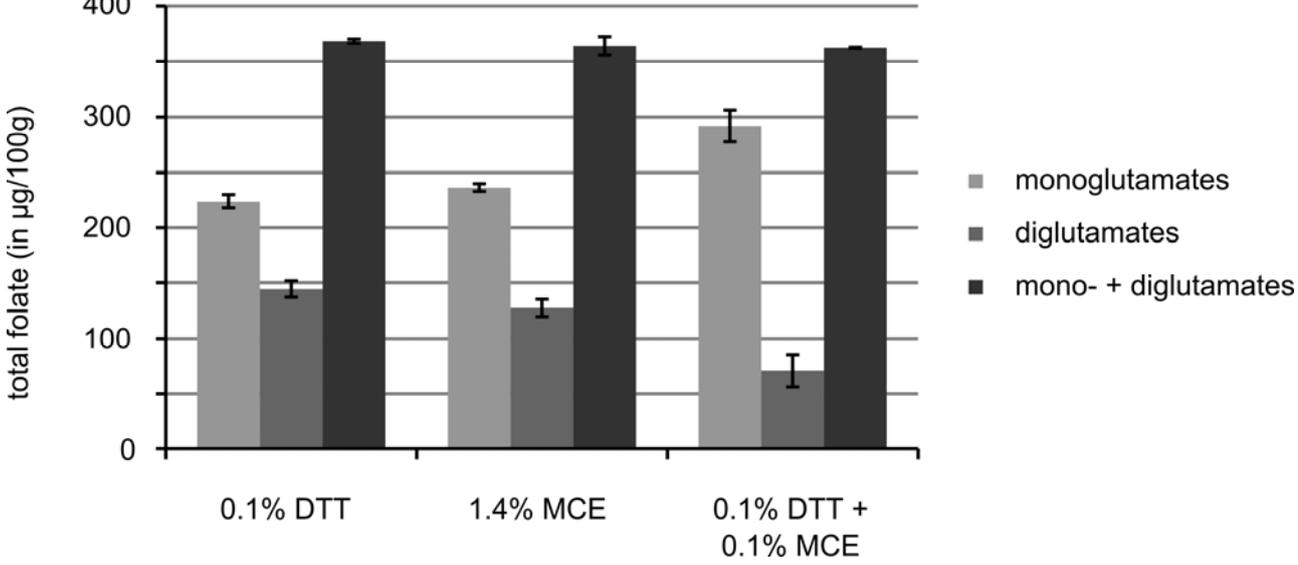
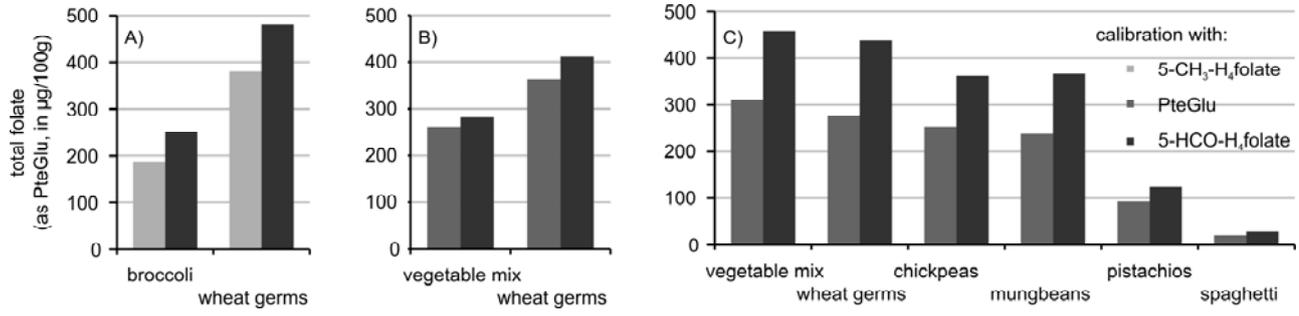


Figure 4



1 Figure 5

2



3