

1 ABSTRACT

2 New stable isotope dilution assays were developed for the simultaneous quantitation
3 of [$^{13}\text{C}_5$]-labelled and unlabelled 5-methyltetrahydrofolic acid, 5-formyltetrahydrofolic
4 acid, folic acid along with unlabelled tetrahydrofolic acid and 10-formylfolic acid in
5 clinical samples deriving from human bioavailability studies, plasma, ileostomy
6 samples and food. The methods were based on clean up by strong anion exchange
7 followed by LC-MS/MS detection. Deuterated analogues of the folates were applied
8 as the internal standards in the stable isotope dilution assays. Assay sensitivity was
9 sufficient to detect all relevant folates in the respective samples as their limits of
10 detection were below 0.62 nmol/L in plasma and below 0.73 $\mu\text{g}/100\text{g}$ in food or
11 ileostomy samples.

12 Quantification of the [$^{13}\text{C}_5$]-label in clinical samples offers the perspective to
13 differentiate between folate from endogenous body pools and the administered dose
14 when executing bioavailability trials.

15

1 INTRODUCTION

2 Since their discovery as a group of vitamins in 1941 [1], the dietary intake and
3 requirements of folates are still under discussion and, despite almost seven decades
4 of research, continuously new perceptions are attained. Basically, folates play a
5 crucial role as coenzymes in the metabolism of one-carbon groups and their intake in
6 Europe is considered to be below the dietary recommendations. In this context, folate
7 deficiency is accepted to increase the risk of neural tube defects [2] and is suspected
8 of being associated with the development of certain forms of cancer [3], Alzheimer's
9 disease [4] and cardiovascular disease [5]. Therefore, accurate methods of folate
10 analysis either in foods or in clinical samples are inevitable, in order to study folate
11 metabolism and absorption kinetics.

12 Folate analysis is demanding due to the high number of vitamers, their occurrence in
13 only trace amounts and their susceptibility towards light and oxygen. Stable isotope
14 dilution assays (SIDA) using stable isotope-labelled folates and LC-MS/MS detection
15 have proven their superiority over conventional methods in analysis of clinical and
16 food samples [6]. For dietary recommendations, knowledge on food folate content is
17 as essential as on folate bioavailability. However, since decades the extent of the
18 latter is controversially discussed due to different study designs and analytical
19 methods.

20 Short-term folate absorption is often assessed using plasma folate concentrations
21 and biokinetic assays, such as the area-under-the-curve (AUC) method [7], or urinary
22 folate excretion [8]. Another approach is to quantify non-absorbed folate in human
23 faeces [9] or ileostomal effluent, *i.e.* effluent of the small intestine, of ileostomy
24 volunteers [10] for calculation of the absorption.

25 For differentiation of endogenous folate deriving from body stores and folate from a
26 test dose, stable isotope-labelled folates have been used, on the one hand, as

1 tracers, and, on the other hand, as analytical internal standards (IS), with labels
2 involving deuterium, ^{13}C or ^{15}N . The first to use folate isotopologues as tracers and
3 as IS were Wright et al. [11], who measured [$^{13}\text{C}_6$]-labelled, [$^{15}\text{N}_{1-7}$]-labelled and
4 unlabelled 5-methyltetrahydrofolic acid (5- CH_3 - H_4 folate) in the single ion monitoring
5 mode of an LC-MS using [$^2\text{H}_2$]-folic acid ([$^2\text{H}_2$]-PteGlu) as IS. However, quantitation
6 was hampered by spectral overlap of [$^{15}\text{N}_{1-7}$]-5- CH_3 - H_4 folate with, on the one hand,
7 [$^{13}\text{C}_6$]-5- CH_3 - H_4 folate, and with, on the other hand, unlabelled 5- CH_3 - H_4 folate in
8 single stage LC-MS. Moreover, the use of a structurally different IS, such as [$^2\text{H}_2$]-
9 PteGlu, may decrease accuracy by ion suppression. A more accurate method has
10 been reported recently by Melse-Boonstra et al. [12], who measured [$^{13}\text{C}_6$]-labelled
11 along with [$^{13}\text{C}_{11}$]-labelled 5- CH_3 - H_4 folate as tracer isotopologues and simultaneously
12 quantified unlabelled 5- CH_3 - H_4 folate by using [$^{13}\text{C}_5$]-5- CH_3 - H_4 folate as the IS. In the
13 latter study, spectral overlaps of 5- CH_3 - H_4 folate isotopologues were avoided by
14 labelling of different moiety of the target molecule and their differentiation by LC-
15 MS/MS. However, this investigation was restricted to plasma 5- CH_3 - H_4 folate without
16 an application to food or other clinical samples.

17 From the latter study it can be seen, that for the simultaneous use as IS and as
18 tracer, different labels and differentiation of these isotopologues from each other and
19 from the unlabelled folates is essential. Besides commercially available [$^{13}\text{C}_5$]-
20 labelled folates, fourfold deuterated folates have been extensively used in folate
21 quantitation [13]. Therefore, it appeared straightforward to use [^{13}C]-labelled
22 isotopologues as tracers in new studies on folate bioavailability and deuterated
23 analogues as IS for subsequent folate quantitation in clinical and food samples
24 deriving from the studies. .

25 The present study was conducted to develop a method for simultaneous quantitation
26 of these differently labelled isotopologues in clinical samples deriving from a human

1 ileostomy study on folate bioavailability. Ileostomy volunteers lack colon and its
2 microflora, which might affect folate excretion. Therefore, non-absorbed folate can be
3 estimated by quantification of post-dose folate excretion in ileostomal effluent
4 samples. During this study, food plasma and ileostomal effluent samples were taken
5 to determine the exact amount of [$^{13}\text{C}_5$]labelled folate in the test dose and the amount
6 of non-absorbed folate dose, respectively. Thereafter, the amount of absorbed folate
7 dose was calculated from the difference of the latter amounts. The assessment of
8 folate bioavailability was based on the comparison of absorbed and non-absorbed
9 folate.

10

11 MATERIALS AND METHODS

12 **Materials**

13

14 The following chemicals were obtained from the sources given in parentheses: Acetic
15 acid p.a. (100 %), acetonitrile LiChrosolv, formic acid p.a. (98–100 %), n-hexan
16 LiChrosolv, hydrochloric acid fuming 37 % LiChrosolv, methanol LiChrosolv,
17 potassium dihydrogen phosphate, sodium acetate trihydrate p.a., disodium hydrogen
18 phosphate p.a., sodium hydroxide p.a., water for chromatography (Merck KGaA,
19 Darmstadt, Germany), α -Amylase, Type II-A, from *Bacillus* species, chicken
20 pancreas (CP), folic acid (PteGlu), 2-(N-morpholino)ethanesulfonic acid (MES)
21 monohydrate minimum 99.5 % titration, potassium dihydrogen phosphate dihydrate,
22 protease Type XIV bacterial from *Streptomyces griseus*, sodiumphosphate dibasic
23 dihydrate (Sigma-Aldrich Chemie GmbH, Steinheim, Germany), L(+)-ascorbic acid
24 (VWR International GmbH, Darmstadt, Germany), sodium chloride (Mallinckrodt
25 Baker B.V., AA Deventer, The Netherlands), β -mercaptoethanol (MCE) molecular
26 biology grade (AppliChem GmbH, Darmstadt, Germany), rat serum azide free

1 (BIOZOL Diagnostica Vertrieb GmbH, Eching, Germany), H₄folate, 5-CH₃-H₄folate,
2 10-HCO-PteGlu, 5-HCO-H₄folate (Schircks, Jona, Switzerland), [¹³C₅]-5-CH₃-
3 H₄folate, [¹³C₅]-5-HCO-H₄folate, [¹³C₅]-PteGlu (Merck Eprova AG, Schaffhausen,
4 Switzerland), [²H₄]-H₄folate, [²H₄]-5-CH₃-H₄folate, [²H₄]-10-HCO-PteGlu, [²H₄]-5-HCO-
5 H₄folate, [²H₄]-PteGlu [14].

6

7 **Solutions and standards**

8

9 *Phosphate buffer* consisted of aqueous Na₂HPO₄*2H₂O (0.1 mol/L) adjusted to pH
10 7.4 by 0.1 mol/L KH₂PO₄ solution.

11 *MES buffer (Extraction buffer for plasma, ileostomal effluent and food samples)*
12 contained 0.02 mol/L MES monohydrate, 2% ascorbic acid and 1.4% MCE at pH 5.0.

13 *Equilibration buffer (for SPE of plasma samples)* consisted of aqueous KH₂PO₄ (3
14 mmol/L) adjusted to pH 7.0 by aqueous Na₂HPO₄ (6 mmol/L) and contained 0.2%
15 MCE.

16 *Equilibration buffer (for SPE of ileostomal effluent and food samples)* contained 3
17 mmol/L KH₂PO₄, 6 mmol/L Na₂HPO₄ and 0.2% MCE at pH 7.5.

18 *0.1 mol/L Phosphate buffer (0.2 mol/L MCE)* was prepared by adjusting an aqueous
19 Na₂HPO₄*2H₂O solution (0.1 mol/L) to pH 7.0 with a 0.1 mol/L KH₂PO₄ solution
20 followed by addition of 0.2 mol/L MCE. The buffers described above were prepared
21 on day of use.

22 *Chicken pancreas (CP)* was solved in phosphate buffer (0.1 mol/L, containing 1%
23 ascorbic acid, pH 7.0) at a concentration of 0.2 mg/mL and stirred for 15 minutes.

24 The solution was kept at -20 °C until day of use.

1 *Elution buffer (for SPE of plasma, ileostomal effluent and food samples)* contained
2 5% sodium chloride, 1% ascorbic acid, 0.1 mol/L sodium acetate trihydrate and 0.1%
3 MCE. The buffer was kept chilled and used for maximum 4 weeks.

4 [²H₄]-labelled folate vitamers (H₄folate, 5-CH₃-H₄folate, 10-HCO-PteGlu, 5-HCO-
5 H₄folate, PteGlu) were used as IS.

6 *IS stock solutions* contained 40 – 350 µg [²H₄]-labelled folate/mL MES buffer.

7 *IS working solutions* were prepared by dilution of the stock solutions with MES buffer.

8 IS working solutions for plasma quantitation contained 25 – 450 ng [²H₄]-labelled
9 folate/mL MES buffer, standards for ileostomy effluent and food quantitations
10 contained 250 – 2000 ng [²H₄]-labelled folate/mL MES buffer. All IS solutions were
11 stored in amber glass bottles at -20 °C.

12 To determine the exact concentrations of the used IS solutions a response mixture
13 consisting of IS solution and a solution of the corresponding unlabelled folate
14 derivative in 0.1 mol/L phosphate buffer (0.2 mol/L MCE) or HCl (0.1 mol/L) for 10-
15 HCO-PteGlu was analysed with every batch of samples. Concentrations of solutions
16 of unlabelled folates were determined by spectrophotometry using a SPECORD 50
17 (analytikjena, Jena, Germany; H₄folate: $\lambda = 299$ nm, $\epsilon = 27.71$ L/(mmol*cm); 5-CH₃-
18 H₄folate: $\lambda = 290$ nm, $\epsilon = 23.71$ L/(mmol*cm), 10-HCO-PteGlu: $\lambda = 252$ nm, $\epsilon = 24.87$
19 L/(mmol*cm), 5-HCO-H₄folate: $\lambda = 288$ nm, $\epsilon = 23.41$ L/(mmol*cm), PteGlu: $\lambda = 282$
20 nm, $\epsilon = 27.61$ L/(mmol*cm)) prior to preparation of the response mixture.
21 Concentration of IS working solution was subsequently determined by LC-MS/MS
22 analysis of the response mixture considering the corresponding response curve (see
23 below) and the amount of unlabelled folate in the mixture in a manner similar to that
24 for determination of sample folate concentration.

25

26 **Clinical samples**

1
2 Plasma and ileostomal effluent samples were obtained from volunteers from a
3 bioavailability trial [15]. The study protocol was approved by the Regional Ethical
4 Review Board in Uppsala, Sweden.

5 Plasma samples derived from blood samples collected in EDTA tubes (BD
6 Vacutainer™, Belliver Industrial Estate, Plymouth, UK) and were stored at -20 °C
7 until analysis.

8 Pooled ileostomal effluent samples were stored at -20° C until analysis.

9

10 **Food samples**

11 Wholemeal bread (8% fibre) fortified with $441 \pm 54 \mu\text{g}$ (6S)- $^{13}\text{C}_5$ -5-CH₃-H₄folate/100 g
12 was baked at the Lantmännen test bakery (Järna, Sweden) as previously described
13 [15] immediately frozen and stored at -20°C. Folate content in the bread was
14 quantified before and after the trial.

15

16 **Precipitation of plasma protein**

17 To prevent clotting of plasma protein on the HPLC column, precipitation and removal
18 of plasma protein had to be accomplished during extraction. Either acetonitrile,
19 methanol or 0.1% formic acid in these solvents are considered good precipitation
20 reagents.

21 It was suspected that folates were not stable in organic solvents, as a previous report
22 showed 12% acetonitrile to be the upper limit to prevent folate degradation [16].

23 Therefore, folate stability in methanol was investigated in different buffers containing
24 12% acetonitrile, 1% ascorbic acid, 0.1% MCE and 10%, 25% and 50% methanol,
25 respectively, on day 1 and day 14 after preparation. To stabilize folates during protein
26 precipitation and to avoid an additional step in the procedure, the extraction buffer

1 composition was modified based on the results of investigations on folate stability in
2 acetonitrile and methanol. Modification of elution buffer composition to precipitate
3 plasma protein on the SPE cartridge was considered another straightforward attempt
4 to remove plasma proteins. In the course of testing seven different reagents for
5 protein precipitation, plasma samples (n=2, 400 µl) were spiked with 100 µl 10-HCO-
6 PteGlu (0.8 nmol/L) as IS (c.f. Table 1). 10-HCO-PteGlu was chosen as IS as it had
7 not been detected in plasma extracted with this method previously [17].

8

9

10 **Final extraction procedure of plasma samples**

11 Prior to analysis, plasma was thawed at room temperature in the dark. 200 µL
12 plasma, IS solutions ($[^2\text{H}_4]$ -H₄folate, $[^2\text{H}_4]$ -5-CH₃-H₄folate, $[^2\text{H}_4]$ -5-HCO-H₄folate, $[^2\text{H}_4]$ -
13 PteGlu) in amounts commensurate to expected sample folate concentration (aimed
14 peak area ratio of 0.5 - 5 for standard and analyte) and 200 µL extraction buffer were
15 pipetted into a 2 mL Eppendorf plastic tube. A little magnetic stirrer was added, the
16 tube was closed and samples were stirred for 37 minutes at room temperature in the
17 dark to equilibrate analytes and internal standards. Subsequently the magnetic stirrer
18 was removed and 600 µl methanol were added into the Eppendorf tube to precipitate
19 plasma proteins. The tubes were closed, shaken well and centrifuged for 5 minutes at
20 9000 G (Centrifuge 5417C, Eppendorf, Hamburg, Germany). Supernatants were
21 cleaned up by SPE as detailed below.

22

23 **Extraction of ileostomy samples**

24

25 Frozen ileostomy samples (approx. 12 g) were lyophilised in four steps of 24 h each
26 (400 µbar -20°C, 400 µbar -10°C, 400 µbar 5°C, 0 µbar 20°C; in a model P8K-E-54-5

1 lyophilizer, Dieter Piatkowski, Forschungsgeräte, München, Germany) and thereafter
2 stored at -20 °C. For extraction, 0.5 g lyophilisate was weighted into Pyrex bottles. IS
3 solutions ($[^2\text{H}_4]$ -H₄folate, $[^2\text{H}_4]$ -5-CH₃-H₄folate, $[^2\text{H}_4]$ -10-HCO-PteGlu, $[^2\text{H}_4]$ -5-HCO-
4 H₄folate, $[^2\text{H}_4]$ -PteGlu) in amounts commensurate to expected sample folate
5 concentration (aimed peak area ratio of 0.5 - 3 for standard and analyte), 2 mg
6 protease, 10 mL MES buffer, and a magnetic stirrer were added. After stirring for 5
7 minutes, samples were incubated in a shaking waterbath at 37°C for 4 hours.
8 Enzyme inactivation was conducted by boiling at 100°C for 10 minutes. Samples
9 were cooled on ice immediately. 2 mL of CP solution and 150 µL of rat serum were
10 added prior to overnight incubation (14 h) in a shaking waterbath at 37°C. Enzymes
11 were inactivated by boiling at 100°C for 10 minutes, before the samples were cooled
12 on ice and centrifuged with 4000 rpm at 2°C for 25 minutes (Centrifuge CR 4-12,
13 Jouan S.A., St. Herblain, France). Supernatants were cleaned up by SPE as detailed
14 below.

15

16 **Extraction of food samples**

17

18 For extraction, 0.1 g of the sample was weighted into Pyrex bottles. IS solutions
19 ($[^2\text{H}_4]$ -H₄folate, $[^2\text{H}_4]$ -5-CH₃-H₄folate, $[^2\text{H}_4]$ -10-HCO-PteGlu, $[^2\text{H}_4]$ -5-HCO-H₄folate,
20 $[^2\text{H}_4]$ -PteGlu) in amounts commensurate to expected sample folate concentration
21 (aimed peak area ratio of 0.5 - 3 for standard and analyte), 3 mg α-amylase, 10 mL
22 MES buffer, and a magnetic stirrer were added. After stirring for 5 minutes, samples
23 were incubated in a shaking waterbath at 37°C for 2 hours. Subsequently, 2 mg
24 protease were added and extraction and SPE clean up were conducted as described
25 for ileostomy samples.

26

1

2 SPE sample clean up

3

4 Plasma samples were purified on a 12-port vacuum manifold (Visiprep, Supelco,
5 Sigma Aldrich, Steinheim, Germany) equipped with strong anion exchange (SAX)
6 SPE tubes (Discovery® DSC-SAX SPE tube 100 mg 1 mL, Supelco, PA, USA; Strata
7 SPE tube 100 mg 1 mL, Phenomenex, Aschaffenburg, Germany), For extracts of
8 ileostomal effluent and foods, the manifold was equipped with SAX SPE tubes
9 (Discovery® DSC-SAX SPE tube 500 mg 3 mL, Supelco, PA, USA; Strata SPE tube
10 500 mg, 3 mL, Phenomenex, Aschaffenburg, Germany). Columns were activated by
11 2 cartridge volumes of n-hexane, 2 volumes of methanol, and 2 volumes of the
12 corresponding equilibration buffer. Following sample application columns were
13 washed with 2 (plasma) or 5 (ileostomal effluent, food) cartridge volumes of the
14 corresponding equilibration buffer and run dry. Plasma folates were eluted with 0.5
15 mL elution buffer.

16 Two attempts were tested in order to increase assay sensitivity for ileostomy and
17 food samples: (1) reduction of elution buffer volume from 2.0 mL to 1.5 mL and (2)
18 fractionated elution of 1 + 1 mL. Based on these results, ileostomy and food folates
19 were eluted with 1.5 ml elution buffer.

20

21 Liquid chromatography

22

23 Chromatography was performed on a High Performance Liquid Chromatograph
24 (Shimadzu Corporation, Kyoto, Japan). The aqueous part of the mobile phase
25 consisted of 1.0% acetic acid in MQ-water (A), the organic part of 0.1% formic acid in
26 acetonitrile (B). The column effluent was directed to waste for the first seven minutes

1 and after 20 minutes (plasma)/ 30 minutes (ileostomal effluent/ food) during each run
2 to avoid accumulation of salt and other interfering compounds in the MS. Injection
3 volume was 20 μ l.

4 For HPLC separation of plasma folates, a Nucleosil C₁₈ column (150 x 2.00 mm, 3 μ ,
5 100 Å) equipped with a C₁₈ precolumn (Phenomenex, Aschaffenburg, Germany) was
6 used. Analysis was carried out at a constant flow rate of 0.2 mL/min. Gradient elution
7 was achieved by variation of the percentage of eluent B at the following times: 0 min
8 0% B, 2 min 10% B, 17 min 30% B, 19 min 100% B, 22 min 0% B, 23 min 0% B.

9 For HPLC separation of ileostomy and food folates, a Nucleosil C₁₈ column (250 x
10 3.00 mm, 5 μ , 100 Å) equipped with a C₁₈ precolumn (Macherey-Nagel, Düren,
11 Germany) was used. Analysis was carried out at a constant flow rate of 0.3 mL/min.
12 Gradient elution was achieved by variation of the percentage of eluent B at the
13 following times: 0 min 0% B, 2 min 10% B, 25 min 25% B, 27 min 100% B, 30 min
14 100% B, 32 min 0% B.

15 To allow for column equilibration between the runs, a 15 minutes equilibration interval
16 at 100% A was included.

17

18 **Mass spectrometry**

19

20 Mass spectrometry was conducted by using an API 4000 Q-Trap LC-MS/MS system
21 (Applied Biosystems, MDS SCIEX, CA, USA) equipped with a turbo spray ion source
22 and operated in positive ion mode. General operating conditions were chosen as
23 follows: source temperature: 400.0°C, spray voltage: 5500.0 V, collision gas
24 pressure: high, curtain gas: 15 psi, ion source gas 1: 38.0 psi, ion source gas 2: 30.0
25 psi, scan type: MRM. Analyte specific operating conditions were tuned as follows:
26 H₄folate, [²H₄]-H₄folate, PteGlu, [²H₄]-PteGlu, [¹³C₅]-PteGlu: DP = 53 V, EP = 8 V, CE

1 = 32 V, CXP = 7 V; 5-CH₃-H₄folate, [²H₄]-5-CH₃-H₄folate, [¹³C₅]-5-CH₃-H₄folate: DP =
2 71 V, EP = 4 V, CE = 23 V, CXP = 8 V; 10-HCO-PteGlu, [²H₄]-10-HCO-PteGlu: DP =
3 59 V, EP = 9 V, CE = 38 V, CXP = 6 V, 5-HCO-H₄folate, [²H₄]-5-HCO-H₄folate, [¹³C₅]-
4 5-HCO-H₄folate: DP = 54 V, EP = 6 V, CE = 23 V, CXP = 10 V. MS detection was
5 based on the following transitions given in table 1.

6

7 **Calibration (Linearity range of response)**

8

9 Separate response curves were measured for each investigated folate derivative
10 (H₄folate, 5-CH₃-H₄folate, [¹³C₅]-5-CH₃-H₄folate, 10-HCO-PteGlu, 5-HCO-H₄folate,
11 [¹³C₅]-5-HCO-H₄folate, PteGlu and [¹³C₅]-PteGlu) in plasma, ileostomal effluent, and
12 foods to convert area ratios (A(IS)/A(Analyte)) into molar ratios (n(IS)/n(Analyte)) to
13 the respective internal standard ([²H₄]-H₄folate, [²H₄]-5-CH₃-H₄folate, [²H₄]-10-HCO-
14 PteGlu, [²H₄]-5-HCO-H₄folate, [²H₄]-PteGlu). IS and analyte solutions were mixed to
15 different ratios (0.05 to 50, n=6). These mixtures were analyzed by LC-MS/MS.
16 Response curves were determined by plotting the molar ratios against the peak area
17 ratios and application of simple linear regression. Response stability was checked
18 during every LC-MS/MS analysis by running a response sample of a random molar
19 ratio.

20

21 **Precision (Inter- and Intraassay-CV)**

22

23 For investigation of repeatability of plasma, ileostomy and food analysis, intra-assay
24 CV was determined by extraction of one plasma and one bread sample in triplicate
25 according to the corresponding methods.

1 For determination of reproducibility of plasma, ileostomy and food assays, inter-assay
2 CV was determined by extraction of one plasma and one bread sample in triplicate
3 (plasma)/ duplicate (ileostomal effluent/ food) once a week for four weeks each
4 according to the corresponding methods.

5

6 **Limit of Detection (LOD)**

7

8 LOD for plasma, ileostomal effluent, and food analysis were determined by extraction
9 of sample matrices free from folates. The surrogate plasma matrix consisted of 7%
10 lyophilised egg white and 0.06% sunflower oil in 0.9% NaCl solution. As a surrogate
11 matrix for food and ileostomal effluent, 100% starch was used. LC-MS/MS analysis
12 confirmed that the surrogate matrices did not contain any folates. For determination
13 of LODs and LOQs the matrices were spiked (each in triplicate) with the analytes at
14 four different concentration levels starting slightly above the LOD and covering 2
15 orders of concentration magnitude. After addition of the respective labelled internal
16 standards, all samples underwent sample preparation and clean-up as described
17 above and were finally analyzed by LC-MS/MS. LODs and LOQs were derived
18 statistically from the data according to a published method [18].

19

20 **Recovery and Accuracy**

21

22 Matrices devoid of folates (see above) for plasma, ileostomal effluent, and food
23 analysis were spiked with known amounts of all investigated folate derivatives
24 (ranging between 2 to 5 nmol/L (plasma) and 0.6 to 2.5 µg/100g (ileostomal effluent
25 and food, n=3) and subsequently extracted according to the corresponding methods.

26

1 **Data analysis**

2

3 All data analysis was performed using Analyst software (Applied Biosystems/ MDS,
4 Analytical Technologies, Concord, Ontario, Canada), versions 1.4 and 1.5, from raw
5 mass spectral data.

6

7

8 **RESULTS AND DISCUSSION**

9 For the application of differently labelled stable isotopologues in physiological
10 studies, several prerequisites have to be fulfilled: a) the complete equilibration
11 between IS and analyte(s), b) the stability of the label(s), and c) an unequivocal
12 spectral differentiation (for details see the recent review on SIDAs for mycotoxins
13 [19]). For the anticipated, simultaneous use of tracer isotopologues and IS
14 isotopologues, the compliance with the former two prerequisites has been confirmed
15 in previous method validation studies [6, 13]. However, the latter prerequisite of
16 absent spectral overlaps had to be verified for the [$^{13}\text{C}_5$]-, the [$^2\text{H}_4$]-labelled and the
17 unlabelled isotopologues of H_4 folate, 5- CH_3 - H_4 folate, 10-HCO-PteGlu, 5-HCO-
18 H_4 folate, and PteGlu. Fig. and 2 present as an example the structures and MS/MS
19 transitions of 5- CH_3 - H_4 folate isotopologues, respectively. None of the abundant
20 product ions of [$^{13}\text{C}_5$]-5- CH_3 - H_4 folate was identical with those of the IS [$^2\text{H}_4$]-5- CH_3 -
21 H_4 folate. This observation is in good agreement with the loss of the glutamate group
22 in collision induced dissociation (CID), which generated for labellings in the glutamate
23 and the 4-aminobenzoic acid moiety different product ions. In accordance with these
24 considerations, unlabelled 5- CH_3 - H_4 folate shows the same abundant product ion as
25 the tracer [$^{13}\text{C}_5$]-5- CH_3 - H_4 folate, since the label of the latter is lost during CID.
26 However, both can be differentiated by the mass increment of the precursor ions.

1 These results were verified by recording response mixtures of [¹³C₅]-labelled and
2 unlabelled folates with the respective [²H₄]-labelled folates. All response curves were
3 linear ($R^2 = 0.999$ for all curves) for molar ratios of deuterated standards to analytes
4 ranging at least between 0.3 – 4, mostly between 0.2 -20, and could be described by
5 the following response equation:

$$6 \quad A(\text{IS}):A(\text{Analyte}) = R_f * n(\text{IS}):n(\text{Analyte}) + b.$$

7 Values for R_f and b are shown in Tables 2 and 3. Response curve linearity and good
8 agreement of response factors for [¹³C₅]-labelled and unlabelled folate confirmed the
9 absence of any “cross-talk” effects in the applied mass spectrometer.

10

11 **Plasma analysis**

12 In our recent report on folate analysis in plasma, erythrocytes and urine [17], we
13 described method improvements such as increased folate stability in MES buffer and
14 improved cleanup on SAX columns. However, in routine analysis of plasma samples,
15 further problems pertaining to folate stability and lifetime of HPLC columns occurred.

16 To increase folate stability, we added 2-mercapto ethanol also to the elution buffer of
17 SAX cleanup.

18 HPLC column lifetime was improved by precipitation of plasma protein during the
19 extraction procedure. Preliminary studies on folate stability in methanol revealed that
20 over a period of 14 days H₄folate was not stable at -18 °C in a solution containing
21 50% methanol. However, folate stability at -18 °C was better in solutions with
22 methanol concentrations of 10% and 25% after 14 days. Folates were most stable in
23 the solution containing 25% methanol, even for the most labile vitamer, H₄folate. For
24 instance, recovery after 14 days of storage at -18 °C was 104% for the latter. Thus all
25 modified extraction buffers contained 12% acetonitrile and 25% methanol.

1 5-CH₃-H₄folate, the main plasma folate derivative, and 10-HCO-PteGlu (IS) were not
2 detectable in samples analysed including precipitation reagents containing formic
3 acid (Table 2, reagents R1, R2, R3). 5-CH₃-H₄folate peak areas for analysis including
4 R4 and R6 (area counts 16000 arbitrary units) were comparable to the blank (area
5 counts 15000 arbitrary units), for analysis including R5 and R7 5-CH₃-H₄folate peak
6 areas were increased by one third (area counts 20000 arbitrary units). The same
7 trend was observed for the IS, however a double-peak was visible for 10-HCO-
8 PteGlu when protein precipitation was accomplished by R7. Thus this approach had
9 to be excluded from the possible precipitation techniques due to the remaining risk
10 that other folate derivatives could be degraded as well. It was concluded that protein
11 precipitation was achieved most efficiently by 100% methanol. No adverse effect on
12 plasma H₄folate was observed as the sample was kept in methanol only for a few
13 minutes. Additionally, plasma may be a more protective matrix than the buffers used
14 during the stability experiment.

15
16 The sensitivity of the new equipment was further increased by using 1% acetic acid in
17 the mobile phase according to previous reports [20], which resulted in LODs for the
18 single folate vitamers in plasma ranging from 0.25 to 0.62 nmol/L (see Table 3). In
19 comparison with our preceding method [17] the sensitivity was in the same order of
20 magnitude, although a different triple quadrupole mass spectrometer was applied.
21 The recoveries in the present study were determined in blank recombinates after
22 spiking with unlabelled and [¹³C₅]-labelled folates and analysis as detailed before.
23 The obtained recoveries ranged from 94 to 116% and confirmed the method's
24 accuracy. By using [²H₄]-labelled folates as the internal standards, losses were
25 almost completely compensated for. Absolute recoveries from SPE clean-up ranged
26 from 32 to 88 % as detailed previously [17].

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Analysis of food and ileostomal effluent

Our SIDAs for folates in foods underwent several steps of development in the past and in the present study we report further improvements. In principle, the way of analysis includes extraction in MES buffer after addition of deuterated analogues as IS, followed by incubation with amylase and protease and subsequent deconjugation with rat serum and chicken pancreas conjugase. Cleanup is achieved by SAX and final determination is accomplished by LC-MS/MS.

In this study, we added the quantitation of ^{13}C -labelled monoglutamates to that of the unlabelled folates and increased folate stability by addition of 2-mercaptoethanol to the SAX elution buffer. Moreover, we improved assay sensitivity by reducing the elution volume during SPE sample cleanup from 2.0 mL to 1.5 mL, which led to increased peak signals in mean by 25%. A further experiment by fractionated elution of 1 + 1 mL revealed that endogenous folates were eluted predominantly with the first mL, whereas PteGlu appeared in higher concentrations in the second mL of the elution volume. When comparing fractionated elution and elution with 1.5 mL, the latter was the reasonable compromise in terms of sensitivity and work load. The ratios of unlabelled or [$^{13}\text{C}_5$]-labelled folates to deuterated standards, however, did not differ between all elution patterns indicating that no discrimination between the isotopologues occurs as had been reported during affinity chromatography earlier [6].

Ileostomy samples were treated analogously, but had to be lyophilized prior to extraction due to their high water content.

The sensitivity of the assay was demonstrated by LODs ranging between 0.11 to 0.73 $\mu\text{g}/100\text{g}$ for unlabelled and [$^{13}\text{C}_5$]-labelled folates (Table 4).

1 The recoveries (n=3) in surrogate matrix starch for unlabelled and [$^{13}\text{C}_5$]-labelled
2 folates ranged from 92% to 130%.

3

4 **Precision of the new assays for folates**

5 Precision of real sample analysis was evaluated in an intra-assay study of a sample
6 prepared several times within one day and in an inter-assay study of a sample
7 prepared on several days within four weeks.

8 In the intra-assay and inter-assay studies the CVs in plasma ranged from 4 to 15 %
9 and from 4 to 16 %, respectively (Table 3). In foods and ileostomy samples, the CVs
10 in the intra-assay and inter-assay studies ranged from 1 to 6 % and from 8 to 27%,
11 respectively (Table 4)

12

13 **Application of stable isotope dilution assay to clinical samples**

14 The new SIDAs' suitability was tested by quantifying [$^{13}\text{C}_5$]-labelled and unlabelled
15 folates in clinical samples from one individual and in a bread fortified with [$^{13}\text{C}_5$]-5-
16 $\text{CH}_3\text{-H}_4\text{folate}$. The LC-MS/MS chromatograms of the bread analysis (n=4) are shown
17 in Fig. 3. We found the bread to contain 1.2 ± 0.6 , 10.2 ± 4.5 , 8.1 ± 1.2 , $770.9 \pm$
18 109.7 , 10.7 ± 1.9 , 0.7 ± 0.1 , 21.4 ± 1.9 , 2.5 ± 0.4 , 2.1 ± 0.5 , and 0.8 ± 0.3 $\mu\text{g}/100$ g of
19 H_4folate , [$^{13}\text{C}_5$]- H_4folate , 5- $\text{CH}_3\text{-H}_4\text{folate}$, [$^{13}\text{C}_5$]-5- $\text{CH}_3\text{-H}_4\text{folate}$, 10-HCO-PteGlu,
20 [$^{13}\text{C}_5$]-10-HCO-PteGlu, 5-HCO- H_4folate , [$^{13}\text{C}_5$]-5-HCO- H_4folate , PteGlu, and [$^{13}\text{C}_5$]-
21 PteGlu, respectively. The detection of several [$^{13}\text{C}_5$]-labelled folates indicated a
22 conversion of [$^{13}\text{C}_5$]-5- $\text{CH}_3\text{-H}_4\text{folate}$ during bread making. 27.3, and 10.3 nmol/L of 5-
23 $\text{CH}_3\text{-H}_4\text{folate}$ and [$^{13}\text{C}_5$]-5- $\text{CH}_3\text{-H}_4\text{folate}$, respectively, were quantified in a volunteer's
24 post dose plasma sample after consumption of the test bread. In a post dose
25 ileostomal effluent sample were found 5.42 ± 0.93 , 3.37 ± 0.06 , 2.93 ± 0.09 , $6.54 \pm$
26 0.17 , 1.32 ± 0.07 , and 0.18 ± 0.03 $\mu\text{g}/100\text{g}$ of H_4folate , 5- $\text{CH}_3\text{-H}_4\text{folate}$, [$^{13}\text{C}_5$]-5- $\text{CH}_3\text{-}$

1 H₄folate, 10-HCO-PteGlu, 5-HCO-H₄folate, [¹³C₅]-5-HCO-H₄folate, and PteGlu,
2 respectively.

3

4 CONCLUSIONS

5 The SIDAs presented here are the first reported methods to quantify [¹³C₅]-labelled
6 and unlabelled isotopologues of the most important folyl monoglutamates in plasma,
7 foods, and ileostomal effluent samples.

8 This new methodology offers the perspective to open up new prospects on folate
9 retention during food processing and human folate metabolism. In foods, the
10 reactivity of single folates and the development of reaction products can be studied
11 by using [¹³C₅]-labelled folates regardless of the endogenous folate spectrum.
12 Additionally, the monitoring of known degradation products or interconversion
13 products such as 10-HCO-PteGlu or 4-aminobenzoylglutamate or identification of
14 new products arising from processing or storage will be possible.

15 Moreover, ileostomy applications will allow deriving novel information on folate
16 turnover during transition through the gastrointestinal tract. Absorption can be
17 followed specifically using a tracer folate. By quantitation of [¹³C₅]-5-CH₃-H₄folate,
18 bile secretion and reabsorption of dosed folates is observable in the ileostomal
19 effluent. Moreover, the detection of the trace label in plasma enables to assess
20 absorption and distribution of folates and will allow insights in liver metabolism.

21 As our method is also validated for plasma samples, new inroads into folate
22 bioavailability and metabolism research are accessible. In that way, the combination
23 of an isotopic labelling technique and an AUC/ileostomy model offers the perspective
24 to open new insights in the “black box” liver in folate metabolism.

25

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1 **Table 1:** Investigated folate derivatives in clinical and food samples and MS/MS
 2 transitions for their detection
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Vitamers	Precursor ion (MH ⁺) [m/z]	Product ion (MH ⁺ - γ-glutamate) [m/z]
H ₄ folate	446.1	299.2
[² H ₄]-H ₄ folate	450.1	303.2
5-CH ₃ -H ₄ folate	460.0	313.2
[² H ₄]-5-CH ₃ -H ₄ folate	464.0	317.2
[¹³ C ₅]-5-CH ₃ -H ₄ folate	465.0	313.2
10-HCO-PteGlu	470.1	295.2
[² H ₄]-10-HCO-PteGlu	474.1	299.2
5-HCO-H ₄ folate	474.1	327.2
[² H ₄]-5-HCO-H ₄ folate	478.1	331.2
[¹³ C ₅]-5-HCO-H ₄ folate	479.1	327.2
PteGlu	442.1	295.2
[² H ₄]-PteGlu	446.1	299.2
[¹³ C ₅]-PteGlu	447.1	295.2

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1 **Table 2:** Reagents added to plasma for protein precipitation

Reagent number	Final composition in plasma	Additional step ^a
R1	0.1% formic acid in acetonitrile	+
R2	0.1 % formic acid in methanol	+
R3 (modified extraction buffer I)	25% R2, 12% R1, 0.02 mol/L MES, 2% ascorbic acid, 0.1% MCE	-
R4	100% acetonitrile	+
R5	100% methanol	+
R6 (modified extraction buffer II)	25% methanol, 12% acetonitrile, 0.02 mol/L MES, 2% ascorbic acid, 0.1% MCE	-
R7 (modified elution buffer)	40% methanol, 10% acetonitrile, 5% sodium chloride, 1% ascorbic acid, 0.1% MCE, 0.1 mol/L sodium acetate*3H ₂ O	-

2 ^a +: additional step to a procedure without protein precipitation -: no additional step

3 MES 2-(N-morpholino)ethanesulfonic acid; MCE β-mercaptoethanol

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Table 3. Validation data for the new SIDAS for plasma folates

Compound	Linearity range of response	Parameters for response equation :		LOD	LOQ	Recovery	Precision	
		for the ratio $n(\text{IS}):n(\text{Analyte}) = R_f^*$		[nmol/ L] (n=3)	[nmol/ L] (n=3)	[%] (n=3)	[%] (n=3)	
	A(IS):A(Analyte)	A(IS):A(Analyte) + b					Intra assay	Inter assay
		slope R_f	intercept b					
H ₄ folate	0.02 – 5.30	0.370	+0.042	0.25	0.51	116±6	6	
5-CH ₃ -H ₄ folate	0.02 – 4.60	1.231	-0.033	0.62	1.24	113±8	4	4
[¹³ C ₅]-5-CH ₃ -H ₄ folate	0.03 – 7.80	1.324	-0.014	0.28	0.55	103±1	3	8
10-HCO-PteGlu								
5-HCO-H ₄ folate	0.15 – 45.45	0.826	-0.489					
[¹³ C ₅]-5-HCO-H ₄ folate	0.14 – 42.20	0.826	-0.489					
PteGlu	0.01 – 2.00	0.287	-0.013	0.39	0.77	94±11	15	16
[¹³ C ₅]-PteGlu	0.01 – 3.60	0.283	0.000					

Table 4. Validation data for the new SIDAS for ileostomal effluents and foods

Compound	Linearity range	Parameters for response equation :		LOD	LOQ	Recovery	Precision	
	of response for the ratio	$n(\text{IS}):n(\text{Analyte}) = R_f * A(\text{IS}):A(\text{Analyte}) +$ b		[ppb] (n=3)	[ppb] (n=3)	[%] (n=3)	[%] (n=3)	
	A(IS):A(Analyte)	Rf	b				Intra assay	Inter assay
H ₄ folate	0.03 – 4.75	0.341	+0.011	1.98	3.97	99±10	5	14
5-CH ₃ -H ₄ folate	0.02 – 4.50	1.201	+0.018	1.34	2.69	123±8	6	16
[¹³ C ₅]-5-CH ₃ -H ₄ folate	0.04 – 8.35	1.414	+0.005	1.08	2.17	98±8	1	15
10-HCO-PteGlu	0.01 – 1.35	0.156	+0.001	1.06	2.12	101±32	3	16
5-HCO-H ₄ folate	0.03 – 5.35	0.580	+0.038	2.67	5.34	135±3	4	8
[¹³ C ₅]-5-HCO-H ₄ folate	0.04 – 6.70	0.580	+0.038	5.79	11.58	82±5	2	27
PteGlu	0.01 – 6.50	0.388	+0.017	1.98	3.95	130±22	3	16
[¹³ C ₅]-PteGlu	0.01 – 5.90	0.408	0.000	7.28	14.55	100±17	3	17

